

Arabidopsis Metabolome Analysis Using Infusion ESI FT-ICR/MS

メタデータ	言語: eng 出版者: 公開日: 2015-10-15 キーワード (Ja): キーワード (En): 作成者: Motohashi, Reiko, Satou, Masakazu, Myouga, Fumiyoshi, Oikawa, Akira, Ohta, Daisaku メールアドレス: 所属:
URL	http://hdl.handle.net/10297/9163

1 -Category by field: Plant Science > Plant biochemistry > Other compound
2 Category by field: Plant Science > Plant metabolism > Metabolomics
3 Category by organism: Plants > Arabidopsis > Whole plant > Other compound

4 5 ***Arabidopsis* Metabolome Analysis Using Infusion ESI FT-ICR/MS**

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19 **[Abstract]** We made the method for Arabidopsis metabolome analysis based on direct-infusion
20 Fourier transform ion cyclotron resonance mass spectrometry (FT-ICR/MS) (IonSpec). This
21 method was sufficiently applied to metabolic phenotyping of Arabidopsis. This method is simple
22 in that after homogenizing samples, powdered samples are dissolved in extraction solvents
23 (acetone and methanol) to 20% fresh weight/volume. Extracted sample solutions are dried and
24 dissolved in 50% (v/v) acetonitrile. Mass analysis using FT-ICR/MS (IonSpec) is performed in
25 positive and negative ionization operation modes. Mass spectra are acquired over the 100-1000
26 m/z range and accumulated to improve the S/N ratio.

27 28 **Materials and Reagents**

- 29
30 1. 3-week-old Arabidopsis plants
31 2. Liquid nitrogen
32 3. Acetone, (HPLC grade) Wako 014-08681
33 4. Methanol, (HPLC grade) Wako 134-14523
34 5. Nitrogen gas
35 6. 50% (v/v) acetonitrile, (HPLC grade) Wako 018-19853, Use Distilled water to dilute
36 acetonitrile to 50%(v/v).
37 7. Acetic acid, Wako 014-20063
38 8. 28~30% ammonia solution, WAKO 016-03146

39 9. Ultrapure water (recommended but not required)

40

41 **Equipment**

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43 1. An IonSpec Explorer FT-ICR/MS equipped with a 7-tesla actively shielded
44 superconducting magnet. (Ionspec, Sold by Agilent)

45 2. Glass vial and teflon cap, GL Sciences 1030-46716

46 3. Glass pipette

47 4. 0.45 µm filters (PTFE), ADVANTEC DISMIC 13HP045AN

48 5. Heat block

49

50 **Procedure**

51

52 A. Preparation of samples for FT-ICR/MS analysis

53 1. Homogenize samples with liquid nitrogen to create powder samples of whole plants
54 using mortars and pestles.

55 2. Dissolve in extraction solvents to 20% fresh weight/volume. Two extraction solvents,
56 100% acetone and 100% methanol, are used to elute various polar compounds.

57 3. Filter extracted sample solutions through 0.45 µm filters (PTFE).

58 4. Transfer filtered sample solutions to vials.

59 5. Transfer 1 ml of sample solution to a separate vial. Place the sample solution into a draft
60 chamber and apply nitrogen gas using EN1-16 (TAITEC 0076417-000). At 40 °C
61 solvents will evaporate using a heat block.

62 6. Dissolve samples in 50% acetonitrile (0.1 ~ 1 mL depending on sample varieties) and
63 store at -80 °C.

64 7. Dilute samples 1:14 prior to ESI analysis.

65

66 B. FT-ICR/MS analysis

67 8. Perform Mass analysis using an IonSpec Explorer FT-ICR/MS (Ionspec) in positive and
68 negative ionization operation modes. Ions are generated from an ESI source with a
69 fused silica needle of 0.005-inch i.d. (Oikawa et al. 2006). Samples are infused using a
70 Harvard syringe pump mode 22 at a flow rate of 0.5 to 1.0µL/min through a 100 µL
71 Hamilton syringe. Set the potentials on the electrospray emitters to 3.0kV and -0.3kV for
72 the positive and the negative electrosprays, respectively.

73 9. For positive mode, add 99.5% formic acid (HPLC grade, WAKO 063-04192) to extracted
74 sample solutions (step 7) at a final concentration of 0.1% (v/v). Positive mode internal
75 standards: Lidocaine (anaesthetic) m/z 235.18104 (WAKO 120-02691), Prochloraz
76 (agricultural chemical) m/z 376.03863 (WAKO 164-25131), Reserpine (alkaloid sedative

77 drug) m/z 609.28121 (WAKO 184-00691), Bombesin (peptide) m/z 810.41479 (WAKO
78 339-40861).

79 10. For negative mode, add 28–30% ammonia solution (WAKO 016-03146) to extracted
80 sample solutions at a final concentration of 0.1% (v/v).

81 Negative mode internal standards: 2.4-D (plant hormone) m/z 218.96157 (WAKO
82 040-18532), Ampicillin (antibiotic) m/z 348.10180 (WAKO 017-20531), CHAPS
83 8detergent) m/z 613.388865 (WAKO 341-04721), (GluNAc)₄
84 (Tetra-*N*-acetylchitotetraose) m/z 829.32023 (Tokyo Chemical Industry T2910).

85 11. Mass spectra are acquired over the 100-1,000 m/z range and accumulate to improve the
86 S/N ratio. The time period for accumulation depends on the total ion concentration.
87 Analyze peaks using the IonSpec Omega ver.8 software. Proofread m/z of each peak
88 referencing the internal standards. Measure the product ion mass spectra of each
89 sample three times.

90 12. When ion peaks are detected at least twice out of three successive spectral scans, they
91 are subjected to further data processing as ion signals from actual analytes.

92 13. A total of four mass spectral peaks from two different extraction solvents (methanol and
93 acetone) and two ionization operation modes (positive and negative) are aligned using
94 our in-house Java program. (If you are interested in this in-house Java program, please
95 contact Prof. Ohta.)

96

97 C. Normalization of data

98 Apply global normalization to data. To avoid zero division, missing values are filled with 10⁵
99 as a background signal of FT-ICR/MS. Peak intensities are transformed using a logarithmic
100 scale with a factor of 10. Four data matrices are used to apply global normalization.

101 Global normalization calculation methods are as follows:

102 1. Average intensity is calculated by dividing the total signal by the number of detected
103 peaks in each spectrum.

104 2. Average signal is calculated for all spectra in each elution and charge pair.

105 3. Normalization factor is calculated for each spectrum by dividing the average intensity for
106 each spectrum by total average intensity.

107 4. Normalized intensity is calculated by multiplying the raw intensity in each spectrum by
108 the previously calculated normalization factor. Empirical formulas are inferred by the
109 accuracy of the FT-ICR/MS. Because the sample ions become adduct ions to attach
110 protons and sodium ions, etc. on the ESI source, we assumed the following were
111 involved in the detected peaks [M+H]⁺, [M+Na]⁺, [M+K]⁺, [M+H+methanol]⁺,
112 [M+ammonium]⁺ in positive ionization mode and [M-H]⁻ in negative ionization mode.

113 5. Search for candidate compounds using KEGG (<http://www.genome.jp/kegg/>), NIST

(<http://webbook.nist.gov/chemistry/>) and KNApSack (Shinbo *et al.*, 2006).

Notes

1. In order to obtain reliably reproducible data, it is imperative that the plant growth environment be as uniform / consistent as possible. For example depending on light environment the value of a plant chlorophyll fluorescence will fluctuate. The amount of light a plant receives when next to the side light on the growth incubator is completely different from the light it receives when on the center of the shelf. For the reason it is important to shuffle the location of growth plates, etc.

Recipes

1. It is important that FT-ICRMS maintains a steady vacuum. We must also be careful about the size of the sample cone on the end of the ESI spray nozzle. It is also important to make minor adjustments where necessary to the direction of ESI ionspray, in an effort to capture as many ions into ICR cells as possible.
2. FT-ICRMS can have problems when the concentration of plant extraction samples is too high, or when other unwanted substances are included in samples. Regular baking and removal of ions from the chamber inner wall as well as cleaning of the ion source is required.

Acknowledgments

This work was supported by the Ministry of Education, Culture, Sports, Science and Technology (Japan) [Grants-in-Aid for Scientific Research (No.17681022 to R.M.)].

FT-ICR/MS (IonSpec) machine belongs to Osaka Prefecture University, and for this project it was used according to directions as indicated by Professors Ohta and Oikawa of the same university. Data normalization techniques were established by Mr. Satou.

This protocol is modified and appended referencing the original, as featured in Integrated analysis of transcriptome and metabolome of Arabidopsis albino or pale green mutants with disrupted nuclear-encoded chloroplast proteins. Satou M, Enoki H, Oikawa A, Ohta D, Saito K, Hachiya T, Sakakibara H, Kusano M, Fukushima A, Saito K, Kobayashi M, Nagata N, Myouga F, Shinozaki K, Motohashi R. *Plant Mol Biol*. 2014 Jul;85(4-5):411-28

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