

Fluoroimmunoassay of influenza virus using sulfur-doped graphitic carbon nitride quantum dots coupled with Ag<sub>2</sub>S nanocrystals

メタデータ	言語: eng 出版者: 公開日: 2020-07-27 キーワード (Ja): キーワード (En): 作成者: Achadu, Ojodomo J., Lioe, De Xing, Kagawa, Keiichiro, Kawahito, Shoji, Park, Enoch Y. メールアドレス: 所属:
URL	<a href="http://hdl.handle.net/10297/00027561">http://hdl.handle.net/10297/00027561</a>

1 **Fluoroimmunoassay of influenza virus using sulfur doped-graphitic**  
2 **carbon nitride quantum dots coupled with Ag<sub>2</sub>S nanocrystals**

---

3 Ojodomo J. Achadu<sup>a</sup>, De Xing Lioe<sup>b</sup>, Keiichiro Kagawa<sup>b</sup>, Shoji Kawahito<sup>b</sup>, Enoch Y.

4 Park<sup>a,c,\*</sup>

5 <sup>a</sup> *Research Institute of Green Science and Technology, Shizuoka University, 836*  
6 *Ohya Suruga-ku, Shizuoka 422-8529, Japan*

7 <sup>b</sup> *Research Institute of Electronics, Shizuoka University, 3-5-1 Johoku, Nakaku,*  
8 *Hamamatsu, Shizuoka 432-8011, Japan*

9 <sup>c</sup> *Laboratory of Biotechnology, Department of Bioscience, Graduate School of*  
10 *Science and Technology, Shizuoka University, 836 Ohya Suruga-ku, Shizuoka 422-*  
11 *8529, Japan*E-mail:

12  
13 ojodomo.john.achadu@shizuoka.ac.jp (OJA) (ORCID: 0000-0002-5542-3110)

14 lioe@idl.rie.shizuoka.ac.jp (DXL)

15 kagawa@idl.rie.shizuoka.ac.jp (KK)

16 kawahito@idl.rie.shizuoka.ac.jp (SK)

17 park.enoch@shizuoka.ac.jp (EYP) (0000-0002-7840-1424)

---

\* Corresponding author: Research Institute of Green Science and Technology, Shizuoka University, 836 Ohya, Suruga-ku, Shizuoka 422-8529, Japan.  
E-mail address: [park.enoch@shizuoka.ac.jp](mailto:park.enoch@shizuoka.ac.jp) (E.Y. Park). Tel (Fax): +81-54-238-4887

## 20 Abstract

21 A novel sulfur-doped graphitic carbon nitride quantum dots (S-gCNQDs) are  
22 synthesized using a single-source precursor in a one-step solvothermal process. The  
23 S-gCNQDs with a size of  $\sim 5$  nm display a strong green intrinsic fluorescence at  
24 512 nm when excited at 400 nm, with a quantum yield of  $\sim 33\%$  in aqueous  
25 solution. The prepared S-gCNQDs and Ag<sub>2</sub>S nanocrystals were applied as  
26 innovative functional materials to fabricate a biosensor for virus detection based  
27 on the conjugation of specific anti-human influenza A monoclonal antibody to the  
28 S-gCNQDs and Ag<sub>2</sub>S NCs, respectively. In the presence of the influenza A virus, an  
29 interaction between the S-gCNQDs/Ag<sub>2</sub>S-labeled antibody resulted in the  
30 formation of a nanosandwich structure, which is accompanied by the fluorescence  
31 enhancement of the S-gCNQDs. The change in fluorescence intensity is linearly  
32 correlated with the concentration of the influenza A virus (H1N1) in the 10 fg/mL  
33 to 1.0 ng/mL range, with a limit of detection of 5.5 fg/mL. The assay was further  
34 applied to the determination of clinically isolated influenza A virus  
35 (H3N2/Yokohama) mixed with human serum. The obtained limit of detection  
36 was 100 PFU/mL within the detection range of  $10^2 - 5 \times 10^4$  PFU/mL for the  
37 H3N2 virus. The recovery yield was within the range of 97.6 to 98.1%.

38 **Keywords:** Sulfur-doped graphitic carbon nitride QDs, Silver disulfide nanocrystals,  
39 Influenza A virus, Fluoroimmunoassay, Virus immunoassay, localized surface  
40 plasmonic resonance, Nanosandwich complex.

## 41 Introduction

42 Influenza virus emerged as a major epidemic over a decade ago and has  
43 remained a growing challenge to public health that needs responsive and accurate  
44 diagnostic measures to prevent the spread of the virus and promote early  
45 treatment [1]. Therefore, the current gold-standard methods used for influenza  
46 virus immunoassays are enzyme-linked immunosorbent assay (ELISA), real-time  
47 reverse transcription-polymerase chain reaction (RT-qPCR), immunoblotting assays,  
48 and electrochemical sequence-specific genetic detection [2,3]. Although these  
49 methods offer specific advantages and are widely used, nevertheless, they suffer  
50 some drawbacks due to high operating costs and, in some cases, high interference  
51 by complex matrices. It is, therefore, still necessary to implement cost-efficient  
52 methods.

53 The exceptional optoelectrical properties of fluorescent graphitic carbon nitride  
54 quantum dots (gCNQDs) have earned widespread interest for the design of  
55 promising fluorescent probes [4–6]. The synthesis of these carbon-based  
56 nanosized-particles encompasses the “top-down and “bottom-up” approaches.  
57 The latter approach, which is ubiquitously deployed, involves the advanced  
58 chemical treatments by condensation and/or controlled pyrolysis of small and  
59 organically rich molecules such as formamide, urea, diaminomaleonitrile,  
60 melamine, guanidine, dicyandiamide and organic amines [7–13]. Doping and/or  
61 hetero-atom functionalization of carbon-based graphitic QDs has helped to  
62 improve their optical and electronic properties for various applications. For

63 instance, sulfur-doped graphitic carbon nitride ( $gC_3N_4$ ) nanosheets were prepared  
64 by the ball-milling of melamine and sulfur powder for 48 h and subsequently  
65 treated under hydrothermal condition. The obtained nanosheets were deployed  
66 for enhanced electrocatalytic activity in fuel cells [14]. Phenyl-modified  $gC_3N_4$ ,  
67 which showed a stoke shift of  $\sim 185$  nm and a high fluorescence (FL) intensity, was  
68 prepared and used as a probe for thiram, a pesticide [15]. In another reported  
69 study, an electrochemical sensor for ascorbic acid (AA), dopamine and uric acid  
70 was fabricated based on  $gC_3N_4$  modified with graphene oxide [16]. A complex  
71 hybrid involving Cu-Pd nanoparticles was deposited on  $gC_3N_4$  hybrid nanosheets  
72 for the colorimetric detection of glucose, taking advantage of the peroxidase  
73 mimic activity of the graphitic nanostructured composites [17]. In our previous  
74 studies, the facile one-step preparation of  $gCNQDs$  functionalized with thymine  
75 (T- $gCNQDs$ ) and tannic acid (TA- $gCNQDs$ ), respectively, were reported [9, 10].  
76 The strongly fluorescent  $gCNQDs$  derivatives were deployed as highly competitive  
77 probes for  $Hg^{2+}$  and AA. However, innovative materials are still increasingly  
78 needed for the production of sensors/biosensors with real-time and realistic  
79 diagnostic applications. This ambition motivates the interest in the adoption of  
80 simple one-step *in situ* approaches for novel hybrid heteroatom-doped materials  
81 preparation.

82 In this work, the analytical application of novel S- $gCNQDs$  was examined by  
83 the fluoroimmunoassay of the influenza A virus as a test analyte. A  
84 fluoroimmunosensing method was developed by utilizing the metal-induced

85 fluorescence enhancement of S-gCNQDs in the presence of Ag<sub>2</sub>S nanocrystals  
86 (NCs). To develop the practical biosensor for influenza virus, novel S-gCNQDs  
87 and Ag<sub>2</sub>S NCs, as innovative materials, were covalently conjugated to specific anti-  
88 human influenza A monoclonal antibodies for the capture of the target influenza  
89 A virus. An immuno-reaction ensued between the antibody-labeled S-  
90 gCNQDs/Ag<sub>2</sub>S NCs in the presence of the influenza A virus. As a result, an  
91 immunocomplex structure was formed, followed by an enhancement of the  
92 fluorescence of the S-gCNQDs influenced by the proximity of Ag<sub>2</sub>S NCs. The  
93 developed immunoassay is rapid and achieved a sensitive femtogram limit of  
94 detection (LOD) of 5.5 fg/mL for influenza A virus (H1N1). Also, clinically isolated  
95 influenza A virus (H3N2) was quantified down to 100 PFU/mL and shows that the  
96 developed system is highly sensitive compared to some commercially available  
97 rapid diagnostic test kits (with LODs of ~5000 PFU/mL). This system can be  
98 adapted for the versatile detection of other viral antigens by labeling the S-  
99 gCNQDs and Ag<sub>2</sub>S NCs with suitable capture antibodies, which can induce the  
100 needed immunoreactions.

101

## 102 **Experimental**

### 103 **Chemicals and biological reagents/materials**

104 Silver nitrate (AgNO<sub>3</sub>), 3-Mercaptopropionic acid (MPA, 99 %), dimethyl  
105 formamide (DMF), bovine serum albumin (BSA), Rhodamine 6G, N-

106 hydroxysuccinimide (NHS), and N-(3-dimethylaminopropyl)-N'-ethylcarbodiimide  
107 hydrochloride (EDC) were purchased from Sigma-Aldrich (St Louis, USA,  
108 <https://www.sigmaaldrich.com/>). 4-Amino-3-hydrazino-5-mercapto-1, 2, 4 triazole,  
109 Bovine serum albumin (BSA), 4-mercaptobenzoic acid (4-MBA) and NaOH were  
110 supplied by FUJIFILM Wako Pure Chemical Corporation (Osaka, Japan,  
111 <http://ffwk.fujifilm.co.jp/en/index.html>). Diethylene glycol (DEG) was purchased  
112 from Tokyo Chemical Industry (TCI) (Tokyo, Japan,  
113 <https://www.tcichemicals.com/en/jp/>).

114 Anti-human influenza A (H1N1) monoclonal antibody (clone C179), anti-  
115 human influenza A (H3N2) monoclonal antibody (Clone F49) and anti-human  
116 influenza A (H1, H2, H3) monoclonal antibody (Clone C111) which is positive for  
117 both influenza viruses H1N1 and H3N2 were purchased from Takara Bio. Inc  
118 (Kusatsu, Shiga, Japan, <https://www.takara-bio.com/>). Influenza virus A/New  
119 Caledonia (20/99/IVR/116) (H1N1) was purchased from ProSpec-Tany  
120 TechnoGene Ltd. (Rehovot, Israel, <https://www.prospecbio.com/>). Human Serum  
121 from human male AB plasma, USA origin, sterile-filtered was obtained from  
122 Sigma-Aldrich (St Louis, USA, <https://www.sigmaaldrich.com/>), Dengue virus DNA  
123 was supplied by Integrated DNA Technologies (Iowa, USA) ([www.idtdna.com](http://www.idtdna.com)).  
124 Clinically isolated influenza virus A/Yokohama/110/2009 (H3N2) was kindly  
125 provided by Dr. C. Kawakami of Yokohama City Institute of Health, Japan. Goat  
126 anti-rabbit IgG-HRP were purchased from Santa Cruz Biotechnology (Dallas, Texas,  
127 USA, <https://www.scbt.com/home>). Commercial RIDT kit - *QuikNavi Flu 2* was

128 purchased from Denka -Seiken Co. Ltd. (Tokyo, Japan, <http://denka-seiken.jp/jp/>).  
129 For selectivity studies, Prof. K. Morita of Institute of Tropical Medicine, Nagasaki  
130 University, kindly provided the zika virus used in this study. Noro virus-like  
131 particles (NoV-LPs) were prepared in our lab, according to the previously  
132 reported protocol [18]. Hepatitis E virus-like particles (HEV-LPs) were prepared  
133 according to the previous report [19]. All experiments were carried out using high  
134 purity deionized (DI) water (>18 MΩ·cm). All detection/sampling protocol was  
135 carried out according to the standard protocol for influenza virus immunoassays  
136 [20].

137

## 138 Instrumentation

139 Ground state electronic absorption (UV/vis), fluorescence excitation and emission  
140 spectra were recorded on a filter-based multimode microplate reader (Infinite  
141 F200 M; TECAN, Ltd, Männedorf, Switzerland, <https://www.tecan.com/>). Images  
142 of the transmission electron microscope (TEM) were acquired using JEM-2100F  
143 operating at 100 kV (JEOL, Ltd., Tokyo, Japan, <https://www.jeol.co.jp/en/products/detail/JEM-2100.html>). Powder X-ray  
144 diffraction (PXRD) analysis was carried out using a RINT ULTIMA XRD (Rigaku  
145 Co., Tokyo, Japan, <https://www1.rigaku.com/ja>) with  $\alpha$ Ni filter and a Cu-K $\alpha$   
146 source. Data were collected over  $2\theta = 15 - 60^\circ$  at a scan rate of 0.01°/step and 10  
147 s/point. Dynamic light scattering (DLS) analysis was done on a Malvern Zetasizer



149 nanoseries, Nano-ZS90 (Malvern Inst. Ltd., Malvern, UK,  
150 <https://www.malvernpanalytical.com/>). Fourier transform infrared spectroscopy  
151 was performed using FT/IR-6300 with ATR PRO610P-S (JASCO, Tokyo, Japan,  
152 <https://www.jasco.co.jp/>). Raman spectroscopic measurements and surface-  
153 enhanced Raman scattering (SERS) experiment of 4-MBA (Raman reporter) and  
154 the virus-immunocomplex using Ag<sub>2</sub>S NCs as the SERS substrate were carried out  
155 using NRS-7100 Raman Spectrometer with f500 spectrograph (JASCO, Tokyo,  
156 Japan). The measurements were done using a 20× objective lens at 1 % laser  
157 power and 2 s integration time. Fluorescence lifetime imaging microscopy (FLIM)  
158 experiment for lifetime determination was done using a Nikon Eclipse Ti-U  
159 microscope with a Nikon CFI S Plan Fluor ELWD 40× (NA=0.60) and Hamamatsu  
160 C8898 (Wavelength: 374 nm, Pulse width: 74 ps, peak power: 47mW) as the  
161 laser source (Hamamatsu, Japan). The FLIM camera used was a Lab-made camera  
162 with a custom CMOS image sensor. The images were acquired by the FLIM CMOS  
163 camera via the Framelink PCIe card (VCE-CLEX02). The FLIM CMOS camera has a  
164 pixel array of 128 × 128 pixels; each has a four-tap pixel with a pitch of 22.4 μm ×  
165 22.4 μm. The sensor response time is 170ps, measured with a 472 nm laser diode.  
166 Details of the Phasor plots generation are given in Electronic Supporting  
167 Information (ESM). Conjugation of the antibody to the individual nanoparticles  
168 was confirmed by enzyme-linked immunosorbent assay (ELISA) using a microplate  
169 (Model 680; Bio-Rad, Hercules, USA, <https://www.bio-rad.com/en-us/>).

170

## 171 **Synthesis of S-gCNQDs**

172 Sulfur-doped graphitic carbon nitride QDs (S-gCNQDs) were prepared by the  
173 solvothermal treatment of a novel precursor i.e., mercapto-based triazole  
174 compound (**Scheme 1**). Briefly, 4-Amino-3-hydrazino-5-mercapto-1, 2, 4 triazole  
175 (50 mg), and citric acid (5 mg) were dissolved in 10 mL of DMF, and the mixture  
176 was sonicated for 20 min to obtain a suspension. The resulting mixture was  
177 transferred and sealed in a 50 mL Teflon-lined stainless steel autoclave and heated  
178 at 200 °C for 8 h. The autoclave was allowed to cool naturally, and the obtained  
179 product was filtered through a 0.22 µm microporous filter membrane and then  
180 dialyzed for 2 d using a dialysis tubing membrane - MWCO 2.0 kDa to obtain  
181 pure S-gCNQDs solution. The solution was further freeze-dried to get a solid  
182 product.

183 Ag<sub>2</sub>S NCs were synthesized according to procedures reported previously with  
184 some modification [21]. Detailed synthesis procedures are presented in Electronic  
185 Supporting Information (ESI).

186

## 187 **Antibody conjugation process and virus detection**

188 Firstly, anti-human influenza virus A (H1N1) (Clone C179) or (H1, H2, H3) (Clone  
189 C111) monoclonal antibody was conjugated onto the surface of S-gCNQDs or Ag<sub>2</sub>S  
190 NCs via EDC/NHS chemistry. To achieve this, 100 µL of 0.1 M EDC was added to  
191 2 mL (0.1 mg/mL) of S-gCNQDs to activate the carboxylic groups on their surface,

192 and the solution was stirred for 1 h, following by the addition of 100  $\mu$ L of 0.1 M  
193 NHS to the mixture and the stirring continued for another 1 h. Then 5.1  $\mu$ g/mL of  
194 the antibody (prepared in PBS 7.6) was added to the activated S-gCNQDs, and  
195 the resulting mixture was stirred for 8 h at 7  $^{\circ}$ C. Anti-human influenza virus A  
196 (H3N2) monoclonal antibody (Clone F49) was also conjugated to S-gCNQDs for  
197 the detection of the clinically isolated influenza virus A/Yokohama/110/2009  
198 (H3N2). Similar procedures were followed for the antibody conjugation of the  
199 Ag<sub>2</sub>S NCs, but using anti-human influenza virus A (H1, H2, H3) (Clone C111)  
200 monoclonal antibody instead of anti-human influenza virus A (H1N1) (Clone  
201 C179). The obtained antibody-conjugated S-gCNQDs or Ag<sub>2</sub>S NCs were purified  
202 by centrifugation (3000  $\times$  g, 5 min) to remove unbound antibodies followed by  
203 incubation with 5 % BSA (for blocking) to ensure non-specific binding interactions.  
204 Excess BSA was further removed by centrifugation, and the conjugates redissolved  
205 in ultra-pure DI water for further use.

206 Following the antibody conjugation, the virus assay was carried out using  
207 different concentrations of influenza A virus (H1N1) within the range from 1.0  
208 fg/mL to 10 ng/mL. Typically, in a 96-well plate, 100  $\mu$ L of the antibody-  
209 conjugated S-gCNQDs ( 2.0 mg/mL in PBS, pH 7.6) was incubated with 100  $\mu$ L of  
210 each concentration (1, 10, 10<sup>2</sup>, 10<sup>3</sup>, 10<sup>4</sup>, 10<sup>5</sup>, 10<sup>6</sup> fg/mL) of the influenza A/New  
211 Caledonia virus (H1N1) for 2 min. Then 50  $\mu$ L of the antibody-conjugated-Ag<sub>2</sub>S (1  
212 mg/mL in PBS, pH 7.6) was added to the mixture and shaken for 5 min to induce  
213 the virus-mediated nanosandwich complex formation. Then the FL intensity

214 change of the S-gCNQDs in the hybrid sandwich nanostructure was measured at  
215 400 nm excitation wavelength, with maximum emission intensity at 512 nm to  
216 construct a calibration curve.

217 The analysis of clinically isolated influenza A/Yokohama (H3N2) was assayed  
218 using similar procedures upon mixing with human serum. 100  $\mu$ L of the antibody-  
219 conjugated S-gCNQDs (2.0 mg/mL in PBS, pH 7.6) was mixed with 50  $\mu$ L of the  
220 clinical samples dissolved in PBS (pH 7.6) in 40 % diluted human serum and  
221 incubated for 2 min. This solution was followed by the addition of 50  $\mu$ L of the  
222 antibody-conjugated-Ag<sub>2</sub>S NCs (1 mg/mL in PBS, pH 7.6) and then thoroughly  
223 shaken for 5 min and allowed for about further 10 min before the FL signals were  
224 collected at 512 nm emission wavelength upon excitation at 400 nm. The  
225 detected range was within 20 – 50,000 PFU/mL of the clinical samples. All  
226 detection experiments were done in triplicate under a similar procedure and at  
227 optimized conditions (See ESM for details).

228

## 229 **Results and discussion**

### 230 **Choice of materials**

231 The surface functionalization and/or doping of carbon-based QDs with  
232 heteroatoms (S, N, and B) are known to result in substantial improvements in  
233 their optical properties and performance [22–24]. This is because their optical  
234 properties are altered by the introduction of “emissive trap sites” and/or “surface

235 defects,” which influence the radiative recombination of their excitons [25, 26].  
236 This work presents an exceptionally rich novel source precursor for the  
237 preparation of gCNQDs with heteroatoms (S and N) functionality using 4-Amino-  
238 3-hydrazino-5-mercapto-1, 2, 4 triazole. This compound contains a high  
239 percentage of the amino group (NH<sub>2</sub>) (essential for covalent attachments) and N-  
240 atoms linked in a triazole ring with extended C-N and C=S linkages (Scheme 1).  
241 The S, N-derived gCNQDs herein has a significant advantage which is the  
242 connective affinity to interact with Au or Ag-based nanoparticles to form  
243 functional hybrids. In addition to the presence of a planar graphitic structure for  
244 non-covalent  $\pi$  stacking interactions, S-gCNQDs prepared using this precursor  
245 possesses more than one point of attachment for simple surface modifications and  
246 hybrid nano-structuring choices.

247 On the other hand, plasmonic semiconductor nanostructures can act as  
248 nanoantennas when close to fluorophores resulting in changes in their optical  
249 properties. This mostly leads to a favorable effect known as metal-enhanced  
250 fluorescence (MEF) [27–29]. This type of influence has been used in detection  
251 systems where metallic nanostructures regulate the emissions of colloidal quantum  
252 dots [28]. Semiconductor nanocrystals such as Ag<sub>2</sub>S NCs used in this work, are  
253 known to exhibit tunable plasmonic properties in the near-infrared (NIR) region  
254 and are excellent nanostructures for various optical-based signal enhancement  
255 processes such as MEF, SERS, and photocatalysis [28–32]. It is in this light that we  
256 deployed Ag<sub>2</sub>S NCs in this work to push the fluorescence detection sensitivity of S-

257 gCNQDs for the detection of influenza virus via metal-based enhancement of their  
258 fluorescence by controlling the local environment surrounding the S-gCNQDs via  
259 immunoreactions with influenza virus, as a test analyte.

260

### 261 **Synthesis and characterization of novel S-gCNQDs and Ag<sub>2</sub>S nanocrystals**

262 The precursor for the S-doped gCNQDs was 4-amino-3-hydrazino-5-mercapto-1, 2,  
263 4 triazole. This precursor was inspired by its high nitrogen/sulfur content and  
264 relatively low cost. Hence, the solvothermal treatment and condensation of the  
265 heterocyclic triazole ring of the precursor to achieving novel S-gCNQDs follows a  
266 similar formation mechanism as outlined in literature for the synthesis of gCNQDs  
267 using other triazole compounds [7, 8, 12, 13].

268 To characterize the prepared S-gCNQDs, TEM was employed, and the image  
269 acquired shows a quasi-spherical morphology of the S-gCNQDs. They appear to  
270 be monodispersed with sizes typically estimated within the range of 3–5 nm (**Fig.**  
271 **1A**). To supplement the information on the size determination from the TEM  
272 micrograph, the DLS experiment revealed that the average size distribution of the  
273 gCNQDs was ~5.5 nm (**Fig. 1B**). This result is in close agreement with the size  
274 distribution obtained using TEM images, and further indicates sizes typical of  
275 carbon-based QDs [7, 9, 10]. XRD pattern obtained for the S-gCNQDs displayed a  
276 broad diffraction peak at  $2\theta = 27^\circ$ , which is indexed as (002) lattice typical of  
277 graphitic carbon nitrides (g-C<sub>3</sub>N<sub>4</sub>) (**Fig. 2A**) [7, 12, 13]. This diffraction peak

278 corresponds well in intensity and position with the known d-spacing and  
279 disordered nanostructure of g-C<sub>3</sub>N<sub>4</sub>, thus indicating their formation. Fourier  
280 transform infrared (FTIR) spectroscopy further revealed the base structure of the S-  
281 doped gCNQDs derivative. As shown in **Fig. 2B**, the observed vibration at 662  
282 cm<sup>-1</sup> is assigned to the formed heptazine units of the S-gCNQDs. Characteristic  
283 absorption bands of C=N/C-C and C-N stretching modes of the graphitic structure  
284 were observed at 1493, 1387, and 1097 cm<sup>-1</sup>, respectively. At 1628 cm<sup>-1</sup>,  
285 asymmetric vibrations referring to the carboxylic groups introduced by citric acid  
286 as co-precursor is conspicuously observed. Intense broad peak typical of the -N-H  
287 (resulting from the 2<sup>o</sup> or 3<sup>o</sup> amine moieties) and/or hydroxyl (OH) of the  
288 carboxylic group vibrations have been observed around 3715-2985 cm<sup>-1</sup> as well as  
289 characteristic -CH<sub>2</sub> peaks of the triazine rings at 2930 and 2897 cm<sup>-1</sup>. The presence  
290 of the S-atom was confirmed by the appearance of a strong C=S absorption at  
291 1248 cm<sup>-1</sup>, with an accompanying weak vibration of the C-S bond at 859 cm<sup>-1</sup> (**Fig.**  
292 **2B**). The Raman spectra (**Fig. 2C**) of the solvothermally prepared S-gCNQDs  
293 displayed the characteristic bands indexed at ~1345 (D band) and ~1586 cm<sup>-1</sup> (G  
294 band) due to the disordered sp<sup>2</sup> as a result of the C-N linkage and graphitic sp<sup>2</sup>  
295 layer nanostructures, respectively [7–10, 12, 13]. These results confidently show  
296 that S-gCNQDs were successfully prepared using the novel precursor.

297 The ground-state absorption recorded for the prepared S-gCNQDs is  
298 reminiscent of the electronic transition ( $\pi - \pi^*$ ) of the s-triazine units of the carbon  
299 nitride family [33–35]. An intense absorption occurred in the region <500 nm.

300 Therefore, the S-gCNQDs were subjected to different excitation wavelengths (350  
301 to 500 nm) to determine the optimum emission intensity and wavelength. At an  
302 excitation wavelength of 400 nm, the emission intensity was maximum and  
303 occurred at ~512 nm. Next, the evaluation of the relative fluorescence quantum  
304 yield ( $\Phi_F$ ) of the material was calculated to be ~33 % using Rhodamine 6G as the  
305 reference standard (see Electronic Supporting Information for details). The value  
306 of  $\Phi_F$  for the prepared S-gCNQDs (using 4-amino-3-hydrazino-5-mercapto-1, 2, 4  
307 triazole as the source of S and N) is higher than that reported using other source  
308 precursors such as urea (17.9 %), melamine (5.5 %) or formamide (29 %) [7, 36,  
309 37]. It is demonstrated in this work that the use of 4-amino-3-hydrazino-5-  
310 mercapto-1, 2, 4 triazole results in a graphitic QDs with FL emission extending into  
311 the green region of the visible spectrum which are desired for various applications.

312 Further, fluorescence lifetime imaging microscopy (FLIM) experiments were  
313 conducted to determine the lifetime of the prepared S-gCNQDs. This  
314 measurement was carried out using the frequency-domain FL lifetime  
315 determination known as the Phasor approach [38, 39]. This approach, which  
316 differs from time-correlated single-photon counting method, is accomplished by  
317 analyzing the FLIM data in Phasor space by observing the clustering of pixel values  
318 (from images) in some areas of the generated Phasor plots rather than by fitting  
319 the fluorescence decay using exponentials. As is shown in Fig. 3A, the second (\*),  
320 and third (#) peaks from the measured waveform are due to crosstalk between  
321 different taps in the pixel of the acquired images. This is then deconvoluted to



322 obtain the result, **Fig. 3B**. The average value of the two-component decay lifetime  
323 of S-gCNQDs was determined to be 3.19 ns. The FLIM measurement results of S-  
324 gCNQDs and their immunocomplexes in the presence of the target virus are  
325 shown in Fig. S3 in Electronic Supporting Information (ESM).

326 Detailed Ag<sub>2</sub>S NCs characterization data, experimental results demonstrating their  
327 plasmonic/optical properties, and discussion are presented in Electronic Supporting  
328 Information (ESM).

329

### 330 **Sensitive fluoroimmunoassay**

331 Influenza A/New Caledonia virus (H1N1) was detected based on preliminary  
332 studies using Ab-conjugated S-gCNQDs in the presence of Ab-Ag<sub>2</sub>S NCs to form a  
333 nanosandwich complex leading to signal-amplified fluorescence cycles which are  
334 dependent on the concentrations of the target influenza virus A(H1N1) (**Fig. 4A**).  
335 The Influenza A virus (H1N1) detection was initially carried out in DI water, and  
336 the recorded FL intensities are shown in **Fig. 4A**. Also, since the human serum is a  
337 mixture of complex biological compounds as matrices, the influenza A/New  
338 Caledonian (H1N1) virus was assayed in human serum (40 %) to simulate the  
339 conditions close to actual clinical samples. Hence, it was observed that the S-  
340 gCNQDs could respond to the presence of the influenza A virus (H1N1) within the  
341 linear concentration range of 10 fg/mL to 1 ng/mL (**Fig. 4B**). The detection

342 sensitivity of the probe towards the target virus (H1N1) was evaluated by the  
343 construction of a calibration plot using Eqn. 1 [40].

$$344 \frac{\Delta F}{F_0} = 0.1 + K[\text{Virus}] \quad (1)$$

345 where  $\Delta F$  is the difference between FL intensity before ( $F_0$ ) and after ( $F$ ) addition  
346 of virus,  $K$  is the fluorescence enhancement factor.  $[\text{Virus}]$  is the concentration of  
347 the target virus. The expression  $\Delta F/F_0 = (F_{\text{with virus}} - F_0)/F_0$  gives a ratio of the  
348 fluorescence enhancement to the fluorescence signal before virus addition ( $F_0$ ) [41].

349 Overall, the increased concentrations of influenza A virus (H1N1) led to a  
350 proportional enhancement in the FL intensity of S-gCNQD measured in DI water  
351 and human serum, respectively. Interestingly, the sensitivity achieved in both  
352 media showed an only but slight difference, with similar linearity of their  
353 calibration plots (**Fig. 4B**). The analytical figures of merit of the assay were  
354 assessed by calculating the limits of detection (LODs) using the equation  
355 ( $\text{LOD} = 3\delta/K$ ) [42], where the standard deviation ( $\delta$ ) of 10 replicated  
356 measurements ( $n = 10$ ) was taken.  $K$  is the value of the slopes of the linear  
357 calibration plots. The LOD was calculated to be 5.5 fg/ml in DI water and  
358 8.5 fg/ml in human serum, respectively (**Table S1**).

359 Detailed results and discussion of optimization and control studies leading to the  
360 established detection of influenza virus in work are given in the Electronic  
361 Supporting Information (ESM).

362

### 363 **Assay of clinically isolated influenza virus (H3N2)**

364 To detect clinically isolated influenza A virus (H3N2), the S-gCNQDs  
365 modified with the H3N2 monoclonal antibody was employed (instead of H1N1  
366 monoclonal antibody) in the presence of the Ab-Ag<sub>2</sub>S NCs to form a sandwich  
367 nanocomplex. It was observed that the Ab (H3N2)-conjugated S-gCNQDs  
368 responded to the presence of the influenza A virus (H3N2) within the linear  
369 concentration range of 100–50000 PFU/mL (**Fig. 5A**). The detection sensitivity of  
370 the probe towards clinically isolated influenza A virus (H3N2) was evaluated by  
371 the construction of a calibration plot (**Fig. 5A inset**), using Eqn. (1) above. The  
372 calculated LOD was 100 PFU/mL. Also, human serum was spiked with a known  
373 concentration of the clinically isolated influenza A (H3N2). The assay was able to  
374 quantify ~98 % of the virus with good recoveries (**Table S2**). Therefore, the  
375 performance of the developed detection platform for the influenza virus was  
376 further evaluated against a commercial rapid influenza diagnostic kit (RIDT) –  
377 *QuikNavi-Flu 2* (Denka Seiken Co. Ltd, Tokyo, Japan). The clinical samples  
378 assayed using the RIDT showed that influenza virus A/Yokohama/110/2009  
379 (H3N2) samples with  $\geq 1000$  PFU/mL could only be detected. As shown in Fig. S4,  
380 influenza virus A/Yokohama/110/2009 (H3N2) is not detectable at  $< 1000$   
381 PFU/mL by the RIDT kit). Meanwhile, our developed biosensor for the influenza  
382 virus is responsive to the H3N2 viral RNA down to 45 PFU/mL. This result  
383 indicates that our detection system can achieve an upwards of ~10 times more

384 sensitivity when used in place of the commercial RIDT. It should be pointed out,  
385 however, that both detection strategies are, in theory, very different, and the  
386 production of a fast diagnostic kit with our designed system will enhance the  
387 sensitive detection of the influenza virus considerably. Furthermore, clinical  
388 samples containing target influenza virus can be assayed rapidly in  $\sim 15$  min,  
389 which is quite preferable to the clinical used RT-PCR or rapid molecular assays  
390 capable of producing results in approximately 15-30 min, and other molecular  
391 assays capable of detecting RNA or nucleic acid influenza in around 45–80 min [2,  
392 3].

393

#### 394 **Selectivity**

395 The selectivity of the developed assay was probed in the presence of NoV-LPs,  
396 Zika virus, HEV-LPs, and Dengue DNA to establish the extent of the selective  
397 nature of the test. Expectedly, no interferences and/or changes occurred in the  
398 fluorescence signal of S-gCNQDs even to 1 ng/mL of the tested virus and VLPs (**Fig**  
399 **5B**). This observation can be attributed to the specific affinity of the influenza A  
400 virus to bind to the antibody functionalized nanoparticles (S-gCNQDs/Ag<sub>2</sub>S NCs).  
401 This affinity is as a result of the induced changes in the FL intensity of the S-  
402 gCNQDs signals compared to when other non-specific viruses interact with S-  
403 gCNQDs. It is pertinent to state here that this system may present a versatile  
404 detection approach for desired viruses by choosing the appropriate antigen-

405 antibody pair for the sandwich immuno-reactions leading to the sensitive  
406 detection of the mediating virus.

407 However, the proposed assay herein has some limitations. The sensitivity of  
408 this assay is majorly limited by the interference of heavy metals such as  $\text{Hg}^{2+}$  and  
409  $\text{Cd}^{2+}$  ions, which can quench the FL of the S-gCNQDs. Also, the sizes of the S-  
410 gCNQDs and  $\text{Ag}_2\text{S}$  NCs may need to be carefully controlled to achieve the  
411 sandwich formation in the presence of the target virus. Clinical samples matrix  
412 interference may impair the sensitivity of detection. To address this, a magnetic  
413 separation protocol/virus enrichment should be considered.

414

#### 415 **Detection mechanism**

416 In view of the observed FL enhancement in S-gCNQDs when immunocomplex  
417 with influenza virus in the presence of  $\text{Ag}_2\text{S}$  NCs, it was speculated that  
418 aggregation-induced emission and/or metal-plasmonic enhancement effects might  
419 be responsible for the FL signal enhancement. However, results obtained from the  
420 TEM image of the immunocomplex (Fig. S5 in Electronic Supporting Information),  
421 showed no clear evidence of aggregation of nanoparticles of S-gCNQDs or  $\text{Ag}_2\text{S}$   
422 NCs in the presence of the target virus. This outcome diverted our attention to the  
423 possibility of an optical-based enhancement mechanism since the plasmonic  
424 properties of  $\text{Ag}_2\text{S}$  NCs have been reported previously [43, 44]. Raman  
425 measurement and SERS analysis were carried out (detailed results and discussion

426 are presented in Electronic Supporting Information). To elucidate the detection  
427 mechanism of the developed immunoassay by evaluating the plasmonic and/or  
428 optoelectronic coupling effects of  $\text{Ag}_2\text{S}$  NCs, the SERS experiments were carried  
429 out further complement the observed FL enhancement of S-gCNQDs (Fig. 6), the  
430 detailed SERS results and discussion of 4-MBA are shown in Fig. S6 of Electronic  
431 Supporting Information. The results plausibly demonstrated that the coupling  
432 interactions between S-gCNQDs and  $\text{Ag}_2\text{S}$  NCs via plasmonic and/or chemical  
433 interaction resulted in the observed SERS signal enhancement of S-gCNQDs,  
434 similar to report for QDs in the presence of plasmonic nanostructures [45].  
435 Therefore, it is credible to state that the enhanced SERS signals from the S-  
436 gCNQDs/ $\text{Ag}_2\text{S}$  NCs virus-mediated immunocomplex are due to the local optical  
437 field created by electronic interaction between the S-gCNQDs and  $\text{Ag}_2\text{S}$  NCs. This  
438 process might be responsible for the enhancement of the fluorescence signals of  
439 the S-gCNQDs resulting from their proximity to  $\text{Ag}_2\text{S}$  NCs in a sandwiched  
440 network triggered by the target virus (Scheme 2). Overall, these findings and the  
441 control experiments lead us to conclude that  $\text{Ag}_2\text{S}$  NCs influenced the  
442 enhancement of the fluorescence of S-gCNQDs regulated by the target influenza  
443 virus, which resulted in the sensitive detection of the virus.

#### 444 **Conclusion**

445 In this work, a functional and innovative combination of S-doped graphitic QDs  
446 and  $\text{Ag}_2\text{S}$  nanocrystals are deployed for the fluoroimmunoassay of influenza A  
447 virus. The S-doped graphitic QDs, which were prepared using a novel precursor

448 via a rapid one-step solvothermal route, displayed excellent optical properties.  
449 Their functionalization with specific antibodies positive for the target virus,  
450 initiated a virus-regulated interaction between the S-doped graphitic QDs and Ag<sub>2</sub>S  
451 nanocrystals . This resulted in the fluorescence enhancement of S-gCNQDs upon  
452 forming an immunocomplex with the influenza virus in the presence of Ag<sub>2</sub>S  
453 nanocrystals. The fluorescence of S-gCNQDs increased consistently as the  
454 concentration of the virus increased, thus leading to the rapid detection of the  
455 target influenza virus in a highly sensitive and selective manner. The materials  
456 herein present an opportunity to fabricate a novel biosensing platform required  
457 for practical detection of the influenza virus and for testing other potentially  
458 harmful infectious diseases. This assay is rapid, convenient, and versatile as specific  
459 proteins and virus-like particles of clinical interests can conceivably be expanded  
460 when the materials are functionalized with capture antibody of interest.

461

#### 462 **Acknowledgment/Funding**

463 The authors sincerely thank Professor K. Morita of the Institute of Tropical  
464 Medicine, Nagasaki University, for providing the Zika virus. Dr. C. Kawakami of  
465 Yokohama City Institute of Health, Japan, is gratefully acknowledged for  
466 providing the clinically isolated influenza virus A/Yokohama/110/2009 (H3N2).  
467 We also wish to thank Dr. T.C. Li of the National Institute for Infectious Disease of  
468 Japan for providing HEV-LPs. O.J.A and S. K. gratefully acknowledge the Japan  
469 Society for the Promotion of Science (JSPS) for a Postdoctoral Fellowship for

470 Research in Japan (Standard) (Grant No. 19F19348) and the Grant-in-Aid for  
471 Scientific Research (S) (Grant No. 18H05240), respectively.

472

473

474 **Competing interests**

475 None declared.

476 **Ethical approval**

477 This study was approved and carried out according to the guidelines provided by  
478 the Ethics Committee of the Environment and Hygiene Institute in Shizuoka  
479 Prefecture (September 14, 2016).

480

481

482

483

484

485 **References**

486 1. Gatherer G (2009) The 2009 H1N1 influenza outbreak in its historical  
487 context. *J. Clin. Virol.* 45:174-178



- 488 2. Dziąbowska K, Czaczyk E, Nidzworski, D (2018) Detection Methods of  
489 Human and Animal Influenza Virus-Current Trends. *Biosensors*, 8:94.  
490 <https://doi.org/10.3390/bios8040094>
- 491 3. Choi YJ, Kim HJ, Park JS, Oh MH, Nam HS, Kim YB (2010) Evaluation of  
492 new rapid antigen test for detection of pandemic influenza A/H1N1 2009  
493 virus. *J. Clin. Microbiol.* 48:2260–2262.  
494 <https://doi.org/10.1128/JCM.02392-09>
- 495 4. Dong Y, Wang Q, Wu H, Chen Y, Lu CH, Chi Y, Yang HH (2016) Graphitic  
496 carbon nitride materials: sensing, imaging and therapy. *Small* 12:5376–  
497 5393
- 498 5. Ahmad R, Tripathy N, Khosla A, Khan M, Mishra P, Ansari WA, Syed MA,  
499 Hahn YB (2020) Recent Advances in Nanostructured Graphitic Carbon  
500 Nitride as a Sensing Material for Heavy Metal Ions. *J. Electrochem.*  
501 *Soc.* **167**:037519
- 502 6. Cheng Q, He Y, Ge Y, Zhou J, Song G (2018) Ultrasensitive detection of  
503 heparin by exploiting the silver nanoparticle-enhanced fluorescence of  
504 graphitic carbon nitride (g-C<sub>3</sub>N<sub>4</sub>) quantum dots. *Microchim Acta* 185:332–  
505 340
- 506 7. Barman S, Sadhukhan M (2012) Facile bulk production of highly blue  
507 fluorescent graphitic carbon nitride quantum dots and their application as

- 508 highly selective and sensitive sensors for the detection of mercuric and  
509 iodide ions in aqueous media. *J Mater Chem* 22:21832–21837
- 510 8. Tang Y, Su Y, Yang N, Zhang L, Lv Y (2014) Carbon nitride quantum  
511 dots: a novel chemiluminescence system for selective detection of free  
512 chlorine in water, *Anal Chem* 86:4528-4535
- 513 9. Achadu OJ, Revaprasadu N (2019) Tannic acid-derivatized graphitic carbon  
514 nitride quantum dots as an “on-off-on” fluorescent nanoprobe for ascorbic  
515 acid via copper(II) mediation. *Microchim Acta* 186:87-97.  
516 <https://doi.org/10.1007/s00604-018-3203-x>
- 517 10. Achadu OJ, Revaprasadu N (2018) Microwave-assisted synthesis of  
518 thymine-functionalized graphitic carbon nitride quantum dots as fluorescent  
519 nanoprobe for mercury(II). *Microchim Acta* 185:461–469
- 520 11. Xu J, Chen Y, Ma D, Shang JK, Li YX (2017) Simple preparation of  
521 MgO/g-C<sub>3</sub>N<sub>4</sub> catalyst and its application for catalytic synthesis of dimethyl  
522 carbonate via trans-esterification. *Catal Commun* 95:72-76
- 523 12. Li Y, Cai J, Liu F, Yu H, Lin F, Yang H, Lin Y, Li S (2018) Highly crystalline  
524 graphitic carbon nitride quantum dots as a fluorescent nanosensor for  
525 detection of Fe(III) via an inner filter effect. *Microchim Acta* 185:134-140
- 526 13. Liu S, Tian J, Wang L, Luo Y, Sun X (2012) A general strategy for the  
527 production of photoluminescent carbon nitride dots from organic amines

528 and their application as novel peroxidase-like catalysts for colorimetric  
529 detection of H<sub>2</sub>O<sub>2</sub> and glucose. RSC Adv 2:411-413

530 **14.** Xu C, Han Q, Zhao Y, Wang L, Li Y, Qu L (2015) Sulfur-doped graphitic  
531 carbon nitride decorated with graphene quantum dots for an efficient  
532 metal-free electrocatalyst. J Mater Chem A 3:1841-1846

533 **15.** Mei H, Shu H, Lv H, Liu MW, Wang X (2020) Fluorescent assay based on  
534 phenyl-modified g-C<sub>3</sub>N<sub>4</sub> nanosheets for determination of  
535 thiram. Microchim Acta 187:159-167

536 **16.** Zhang L, Liu C, Wang Q, Wang X, Wang S (2020) Electrochemical sensor  
537 based on an electrode modified with porous graphitic carbon nitride  
538 nanosheets (C<sub>3</sub>N<sub>4</sub>) embedded in graphene oxide for simultaneous  
539 determination of ascorbic acid, dopamine and uric acid. Microchim  
540 Acta 187: 149-159. <https://doi.org/10.1007/s00604-019-4081-6>

541 **17.** Darabdhara, G, Boruah PK, Das MR (2019) Colorimetric determination of  
542 glucose in solution and via the use of a paper strip by exploiting the  
543 peroxidase and oxidase mimicking activity of bimetallic Cu-Pd  
544 nanoparticles deposited on reduced graphene oxide, graphitic carbon  
545 nitride, or MoS<sub>2</sub> nanosheets. Microchim Acta 186:13-23.  
546 <https://doi.org/10.1007/s00604-018-3112-z>

547 **18.** Ahmed SR, Takemura K, Li TC, Kitamoto N, Tanaka T, Suzuki T, Park EY  
548 (2017) Size-controlled preparation of peroxidase-like graphene-gold  
549 nanoparticle hybrids for the visible detection of norovirus-like particles.  
550 Biosens. Bioelectron. 87:558-565

- 551 **19.** Li TC, Yamakawa Y, Suzuki K, Tatsumi M, Razak M, Uchida T, Takeda N,  
552 Miyamura T (1997) Expression and self-assembly of empty virus-like  
553 particles of hepatitis E virus. *J Virol* 71:7207-7213.
- 554 **20.** World Health Organization (2017) **WHO information for the molecular**  
555 **detection of influenza viruses** July [accessed 5 June 2020]  
556 [http://www.who.int/influenza/gisrs\\_laboratory/WHO\\_information\\_for\\_the](http://www.who.int/influenza/gisrs_laboratory/WHO_information_for_the_molecular_detection_of_influenza_viruses_20171023_Final.pdf)  
557 [\\_molecular\\_detection\\_of\\_influenza\\_viruses\\_20171023\\_Final.pdf](http://www.who.int/influenza/gisrs_laboratory/WHO_information_for_the_molecular_detection_of_influenza_viruses_20171023_Final.pdf)
- 558 **21.** Jiang P, Zhu CN, Zhang ZL, Tian ZQ, Pang DW (2012) Water-soluble Ag<sub>2</sub>S  
559 quantum dots for near-infrared fluorescence imaging in vivo. *Biomaterials*  
560 33:5130-5135
- 561 **22.** Bankole OM, Achadu OJ, Nyokong T (2017) Nonlinear Interactions of  
562 Zinc Phthalocyanine-Graphene Quantum Dots Nanocomposites:  
563 Investigation of Effects of Surface Functionalization with Heteroatoms. *J*  
564 *Fluoresc* 27:755–766
- 565 **23.** Qu D, Zheng M, Du P, Zhou Y, Zhang L, Li D, Tan H, Zhao Z, Xied Z, Sun  
566 Z (2013) Highly luminescent S, N co-doped graphene quantum dots with  
567 broad visible absorption bands for visible light photocatalysts. *Nanoscale*  
568 5:12272–12277.
- 569
- 570 **24.** Holá K, Sudolská M, Kalytchuk S, Nachtigallová D, Rogach AL, Otyepka M,  
571 Zbořil R (2017) Graphitic Nitrogen Triggers Red Fluorescence in Carbon  
572 Dots. *ACS Nano*. 12:12402-12410

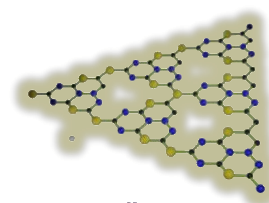
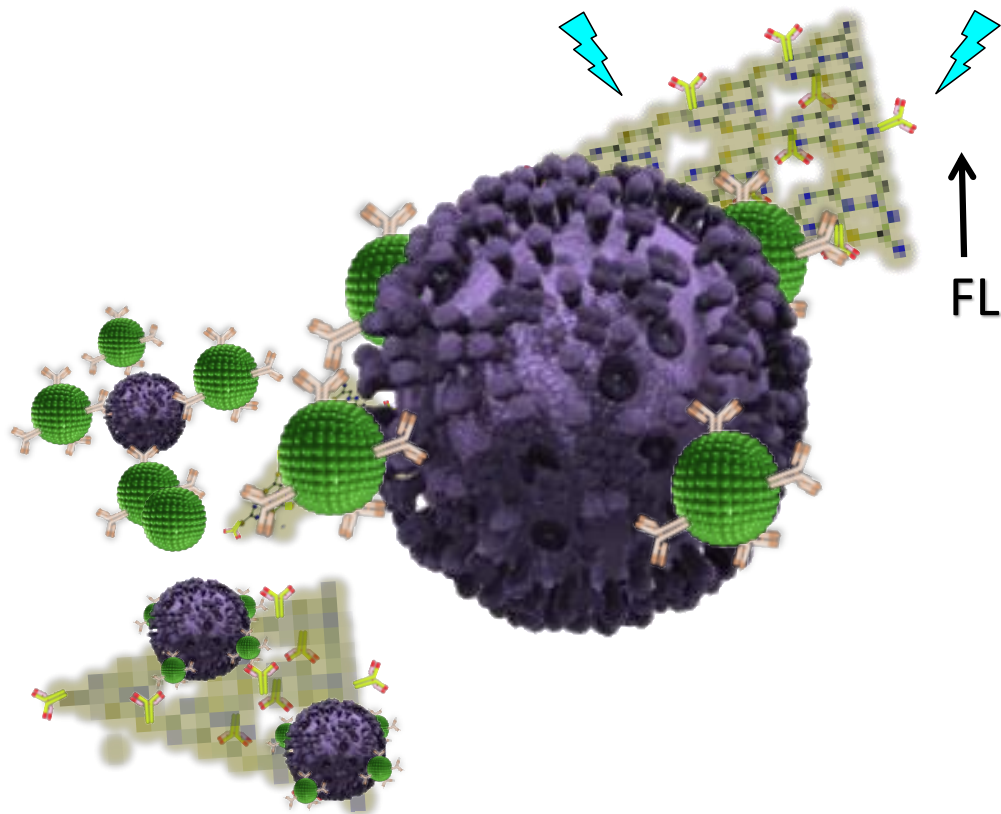
- 573 **25.** Gu SY, Hsieh CT, Gandomi YA, Chang JK, Li, J, Li JL, Zhang HA, Guo Q,  
574 Lau KC, Pandey R (2019) Microwave Growth and Tunable  
575 Photoluminescence of Nitrogen-doped Graphene and Carbon Nitride  
576 Quantum Dot. *J Mater Chem C* 7:5468-5476
- 577 **26.** Wang J, Cao S, Ding Y, Ma F, Lu W, Sun M (2016) Theoretical  
578 Investigations of Optical Origins of Fluorescent Graphene Quantum Dots.  
579 *Sci Rep* 6:24850-24855
- 580 **27.** Lakowicz JR, Ray K, Chowdhury M, Szymanski H, Fu Y, Zhang J,  
581 Nowaczyk K (2008) Plasmon-controlled fluorescence: a new paradigm in  
582 fluorescence spectroscopy *Analyst*, 133: 1308-1346
- 583 **28.** Deng W, Xie F, Baltar HTMCM, Goldys EM (2013) Metal-enhanced  
584 fluorescence in the life sciences: here, now and beyond. *Phys. Chem. Chem.*  
585 *Phys.*, 15:15695-15708
- 586 **29.** Faucheaux JA, Stanton ALD, Jain PK (2014) Plasmon Resonances of  
587 Semiconductor Nanocrystals: Physical Principles and New Opportunities. *J.*  
588 *Phys. Chem. Lett.* 5:976-985. <https://doi.org/10.1021/jz500037k>
- 589 **30.** Lee SH, Nishi H, Tatsuma T (2017) Tunable plasmon resonance of  
590 molybdenum oxide nanoparticles synthesized in non-aqueous media, *Chem.*  
591 *Commun.* **53**:12680-12683

- 592 **31.** Zhang J, Pan Y, Chen Y, Lu H (2018) Plasmonic molybdenum trioxide  
593 quantum dots with noble metal-comparable surface enhanced Raman  
594 scattering, *J. Mater. Chem. C.* 6:2216-2220
- 595 **32.** Zhou Y, Li W, Zhang Q, Yan S, Cao Y, Dong F, Wang F (2017) Non-noble  
596 metal plasmonic photocatalysis in semimetal bismuth films for  
597 photocatalytic NO oxidation. *Phys. Chem. Chem. Phys.* 19:25610-  
598 25616. <https://doi.org/10.1039/C7CP04359G>
- 599 **33.** Achadu OJ, Nyokong T (2017) *In situ* one-pot synthesis of graphitic  
600 carbon nitride quantum dots and its 2, 2, 6, 6-tetramethyl (piperidin-1-yl)  
601 oxyl derivatives as fluorescent nanoprobe for ascorbic acid. *Anal Chim*  
602 *Acta* 991:113–126
- 603 **34.** Zhou J, Yang Y, Zhang CY (2013) A low-temperature solid-phase method  
604 to synthesize highly fluorescent carbon nitride dots with tunable emission.  
605 *Chem Commun* 49:8605-8607
- 606 **35.** Liu S, Wang L, Tian J, Zhai J, Luo Y, Lu W, Sun X (2011) Acid-driven  
607 microwave-assisted production of photoluminescent carbon nitride from  
608 N,N-dimethylformamide. *RSC Adv* 1:951-953
- 609 **36.** Guo J, Lin Y, Huang H, Zhang S, Huang T, Weng W (2017) One-pot  
610 fabrication of fluorescent carbon nitride nanoparticles with high crystallinity  
611 as a highly selective and sensitive sensor for free chlorine. *Sensors and*  
612 *actuators B: chemical* 244:965–971

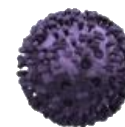
- 613 **37.** Rong M, Lin L, Song X, Wang Y, Zhong Y, Yan J, Feng Y, Zeng X, Chen X  
614 (2015) Fluorescence sensing of chromium (VI) and ascorbic acid using  
615 graphitic carbon nitride nanosheets as a fluorescent switch. *Biosens.*  
616 *Bioelectron* 68:210–217
- 617 **38.** Digman MA, Caiolfa VR, Zamai M, Gratton E (2008) The phasor  
618 approach to fluorescence lifetime imaging analysis. *Biophys J* 94:14–16
- 619 **39.** Chen H, Ma N, Kagawa K, Kawahito S, Digman M, Gratton E (2018)  
620 Wide-field multi-frequency fluorescence lifetime imaging using a two-tap  
621 complementary metal-oxide semiconductor camera with lateral electric  
622 field charge modulators. *J Biophotonics* 12:1–9
- 623 **40.** Lakowicz JR (2009) Principles of fluorescence spectroscopy, Third edn.  
624 Springer, New York, p 243
- 625 **41.** Lee J, Ahmed SR, Oh S, Kim J, Suzuki T, Parmar K, Park S, Lee J, Park EY  
626 (2015) A plasmon-assisted fluoro-immunoassay using gold nanoparticle-  
627 decorated carbon nanotubes for monitoring the influenza virus, *Biosens.*  
628 *Bioelectron.* 64:311-317
- 629 **42.** Lima KMG, Raimundo Jr IM, Pimentel MF (2007) Improving the detection  
630 limits of near infrared spectroscopy in the determination of aromatic  
631 hydrocarbons in water employing a silicone sensing phase, *Sens. Actuators*  
632 *B Chem* 125:229-233

- 633       **43.** Hou X, Zhang X, Yang W, Liu Y, Zhai X (2012) Synthesis of SERS active  
634           Ag<sub>2</sub>S nanocrystals using oleylamine as solvent, reducing agent and stabilizer.  
635           Mater Res Bull 47:2579–2583
- 636       **44.** Fang C, Lee YH, Shao L, Jiang R, Wang J, Xu QH (2013) Correlating the  
637           Plasmonic and Structural Evolutions during the Sulfidation of Silver  
638           Nanocubes. ACS Nano, 7:9354–9365. doi:10.1021/nn404042p
- 639       **45.** Zou F, Zhou H, Van Tan T; Kim J, Koh K, Lee J (2015) Dual-Mode SERS-  
640           Fluorescence Immunoassay Using Graphene Quantum Dot Labeling on  
641           One-Dimensional Aligned Magnetoplasmonic Nanoparticles. ACS Appl.  
642           Mater. Interfaces 7:12168–12175. <https://doi.org/10.1021/acsami.5b02523>  
643  
644





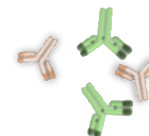
S-gCNQDs



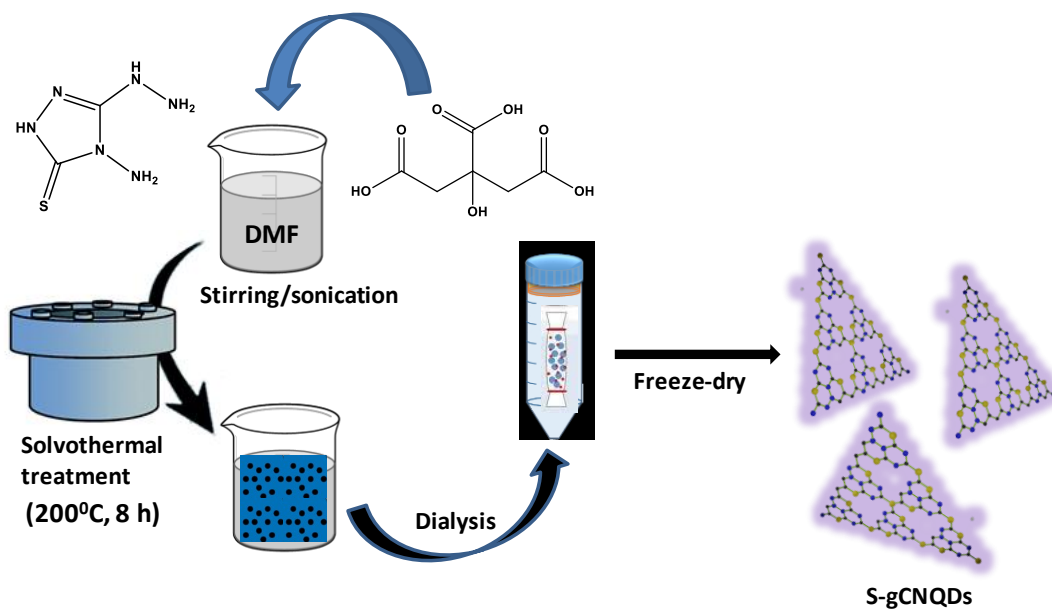
Influenza Virus A/H1N1



Ag<sub>2</sub>S NCs



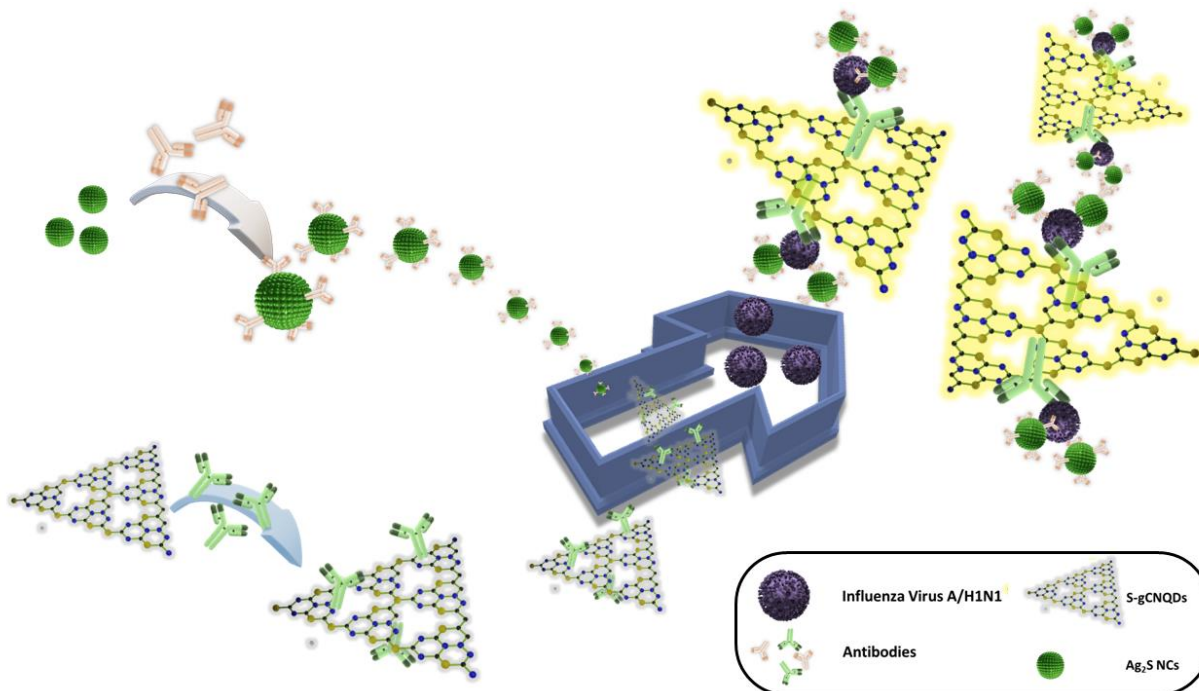
Antibodies



645

646 **Scheme 1.** Synthesis pathway of S-gCNQDs.

647



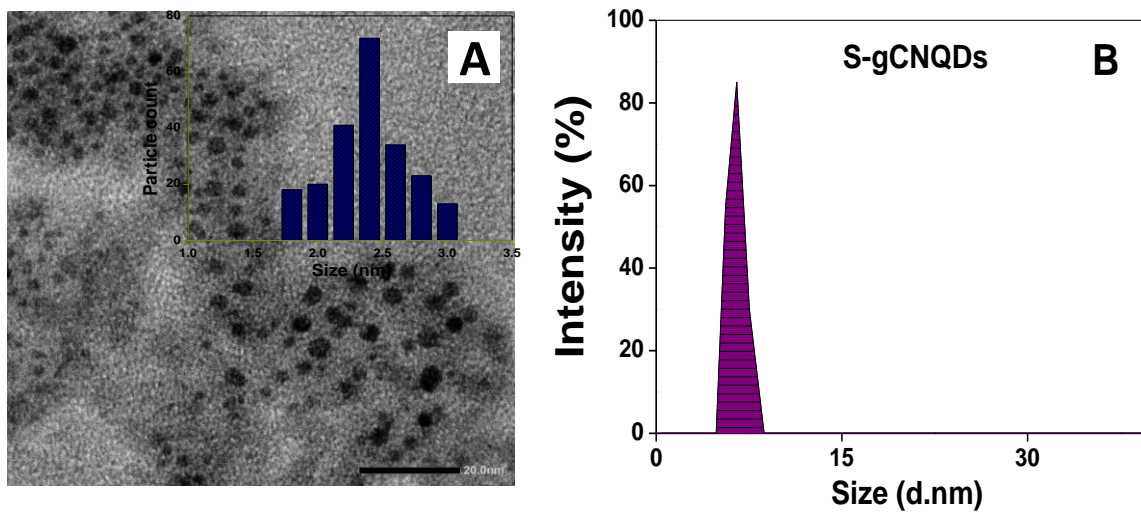
648

649 **Scheme 2.** Schematic representation of the S-gCNQDs and Ag<sub>2</sub>S modification with

650 antibody and the detection protocol by sandwich nanostructure formation in the

651 presence of target influenza A virus.

652

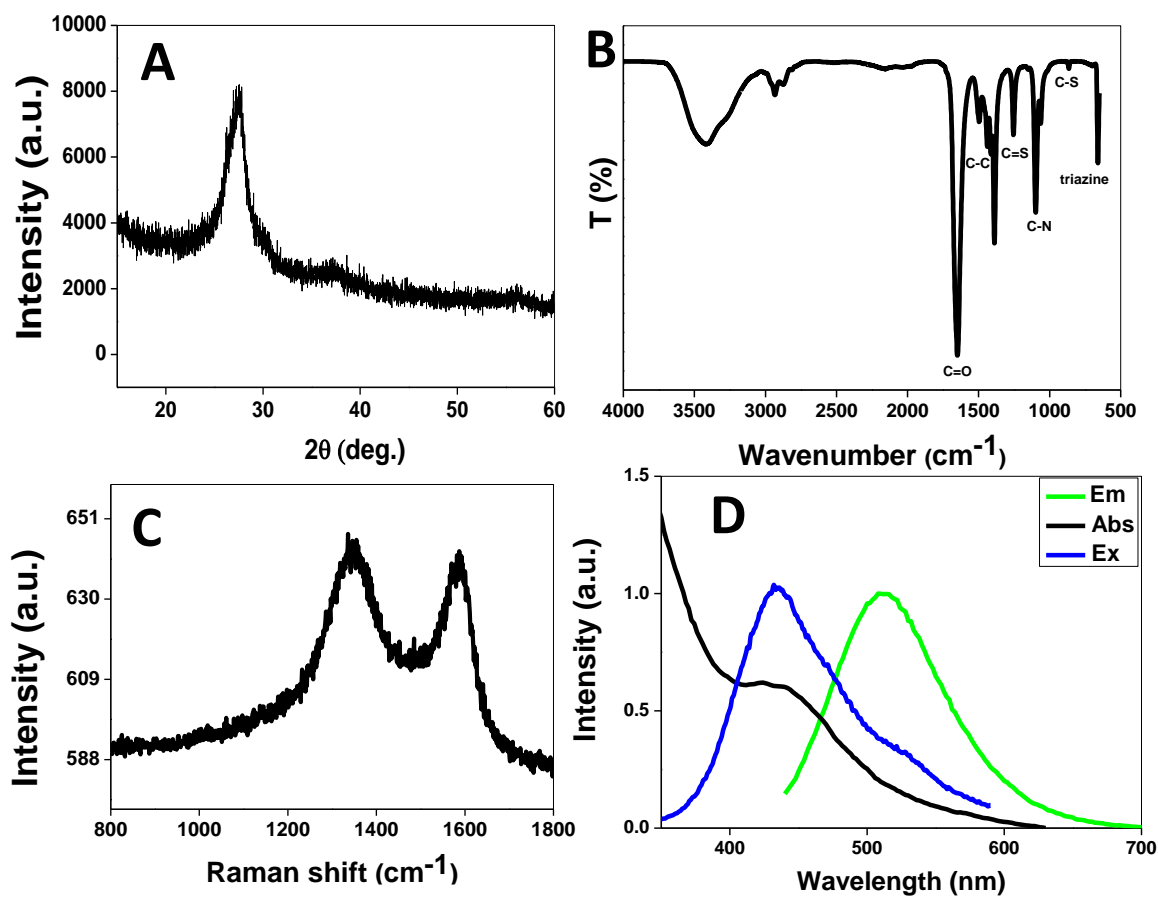


653

654 **Fig. 1.** (A) TEM image of *S*-gCNQDs (inset is the size distribution histogram). (B)

655 DLS graph of *S*-gCNQDs.

656



658

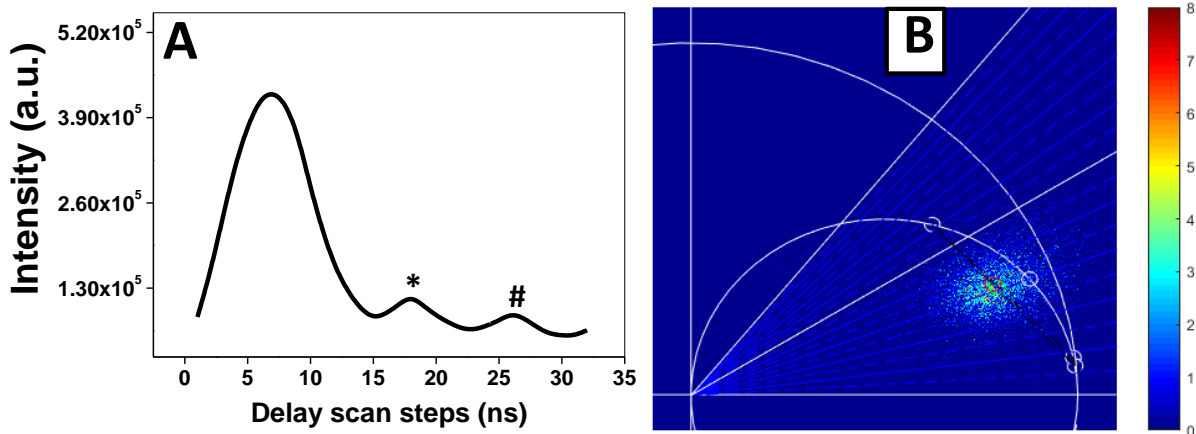
659 **Fig. 2.** Characterization spectra of S-gCNQDs showing (A) XRD pattern (B) FTIR

660 absorptions. (C) Raman spectra and (D) UV-vis, excitation, and emission spectra

661 (Solvent – DI water).  $\lambda_{\text{ex}} = 400 \text{ nm}$ .

662

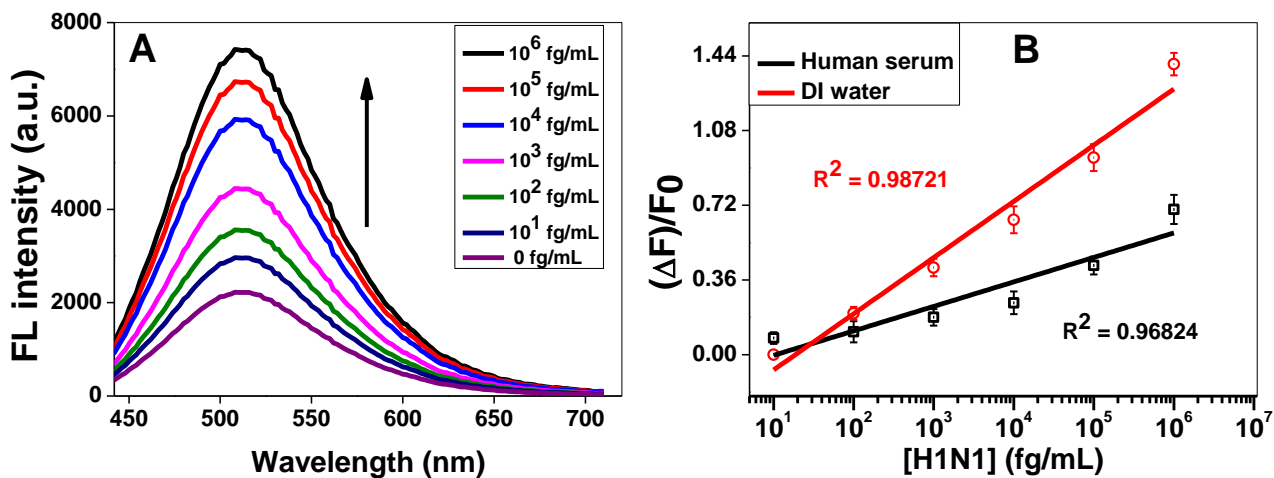
663



664

665 Fig. 3. (A) FL lifetime decay of pristine S-gCNQDs (B) Phasor plot showing the  
666 decay component for the analyzing frequency of 20MHz.

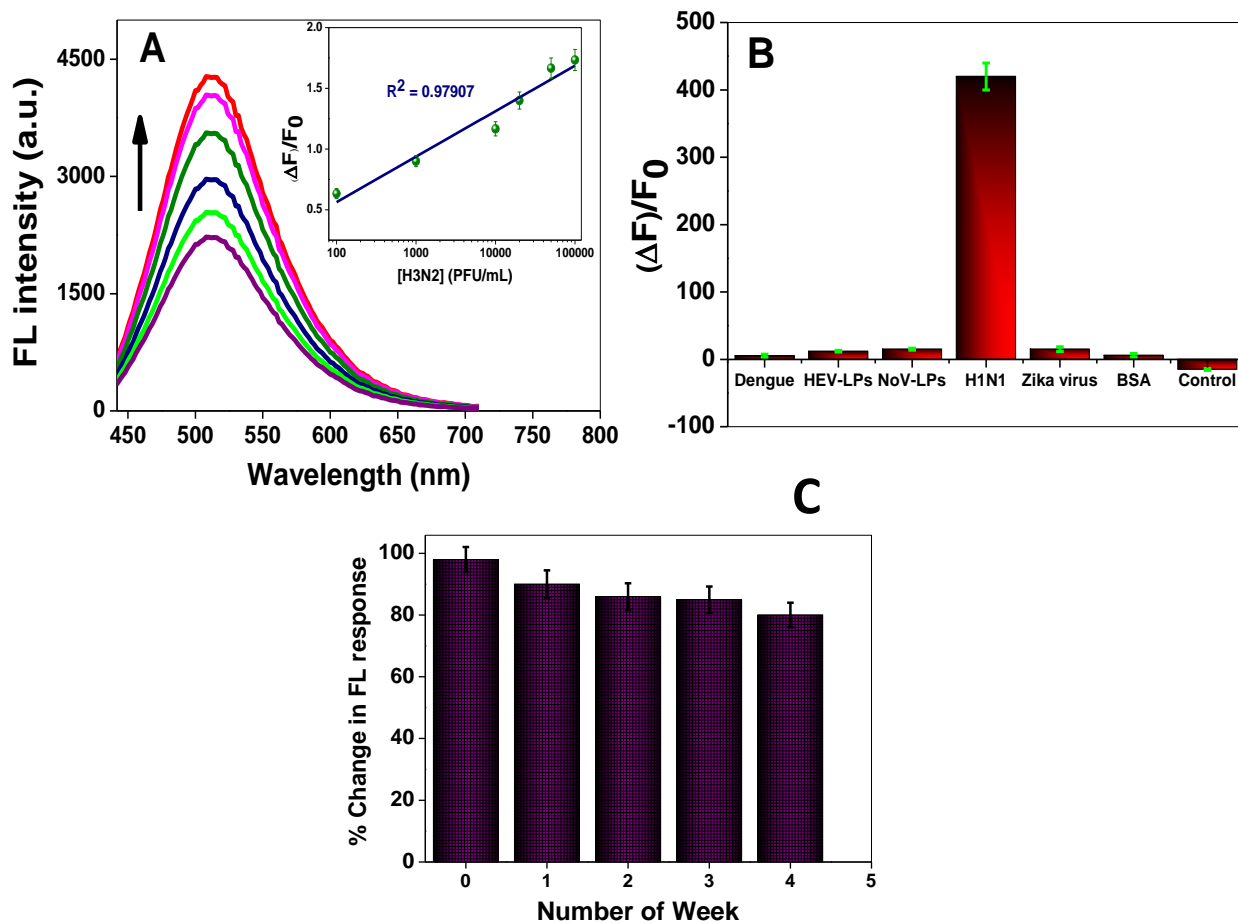
667



674

675 Fig. 4. (A) Detection FL spectra of S-gCNQDs showing enhancement in intensity at  
676 various H1N1 virus concentrations. (B) The corresponding calibration plots  
677 generated in DI water and human serum.  $\lambda_{ex} = 400$  nm.

678



679

680 **Fig. 5.** (A) FL spectra for the detection of clinically isolated influenza A virus  
 681 (H3N2). Inset: the corresponding calibration plot. (B) Plots showing H1N1  
 682 selective assay in the presence of other viruses/V-LPS. (C) Stability test of 5-  
 683 gCNQDs and Ag<sub>2</sub>S NCs with 1 ng/mL of H3N2 virus showing percentage change  
 684 in FL signal response over 4 weeks.  $\lambda_{ex} = 400$  nm.

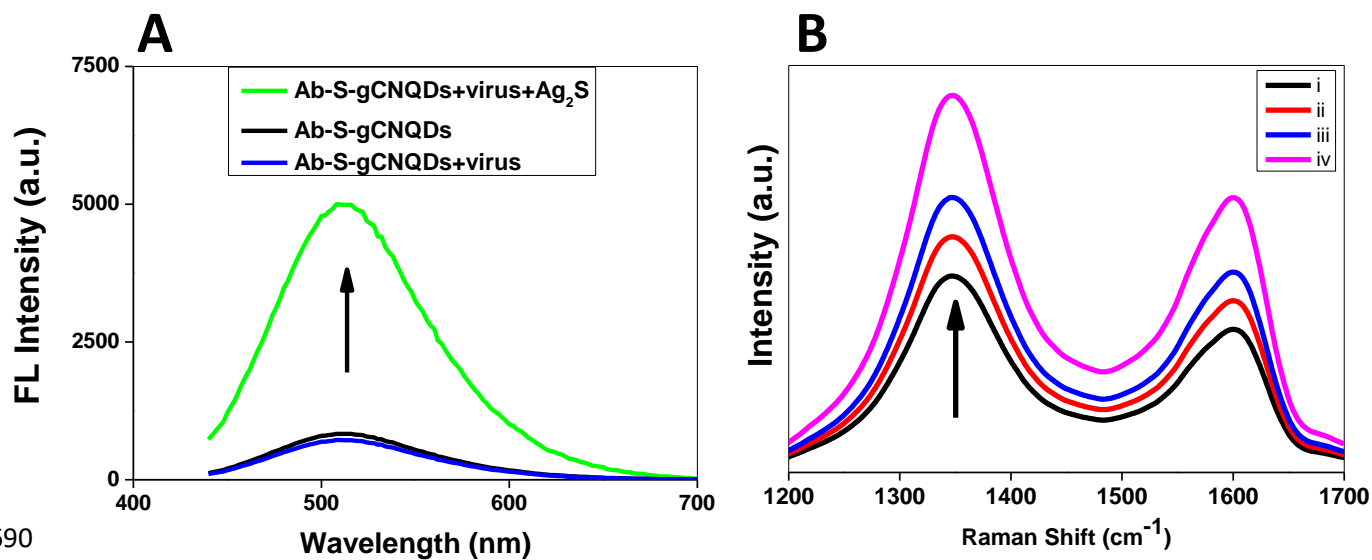
685

686

687

688

689



690

691

692

693 **Fig. 6.** (A) Fluorescence signal enhancement of S-gCNQDs as in the presence of

694 Ag<sub>2</sub>S NCs (B) Corresponding SERS enhancement of S-gCNQDs at different

695 concentrations of Ag<sub>2</sub>S NCs due to local optical coupling effects.

## Electronic Supplementary Material

### Fluoroimmunoassay of influenza virus using sulfur doped-graphitic carbon nitride quantum dots coupled with Ag<sub>2</sub>S nanocrystals

Ojodomo J. Achadu<sup>a</sup>, De Xing Lioe<sup>b</sup>, Keiichiro Kagawa<sup>b</sup>, Shoji Kawahito<sup>b</sup>, Enoch Y.

Park<sup>a,c,\*</sup>

<sup>a</sup> *Research Institute of Green Science and Technology, Shizuoka University, 836 Ohya Suruga-ku, Shizuoka 422-8529, Japan*

<sup>b</sup> *Research Institute of Electronics, Shizuoka University, 3-5-1 Johoku, Nakaku, Hamamatsu, Shizuoka 432-8011, Japan*

<sup>c</sup> *Laboratory of Biotechnology, Department of Bioscience, Graduate School of Science and Technology, Shizuoka University, 836 Ohya Suruga-ku, Shizuoka 422-8529, Japan* E-mail:

ojodomo.john.achadu@shizuoka.ac.jp (OJA)  
lioe@idl.rie.shizuoka.ac.jp (D.X.L.)  
kagawa@idl.rie.shizuoka.ac.jp (K.K.)  
kawahito@idl.rie.shizuoka.ac.jp (S.K.)  
park.enoch@shizuoka.ac.jp (EYP)

---

\* Corresponding author: Research Institute of Green Science and Technology, Shizuoka University, 836 Ohya, Suruga-ku, Shizuoka 422-8529, Japan.  
E-mail address: [park.enoch@shizuoka.ac.jp](mailto:park.enoch@shizuoka.ac.jp) (E.Y. Park). Tel (Fax): +81-54-238-4887



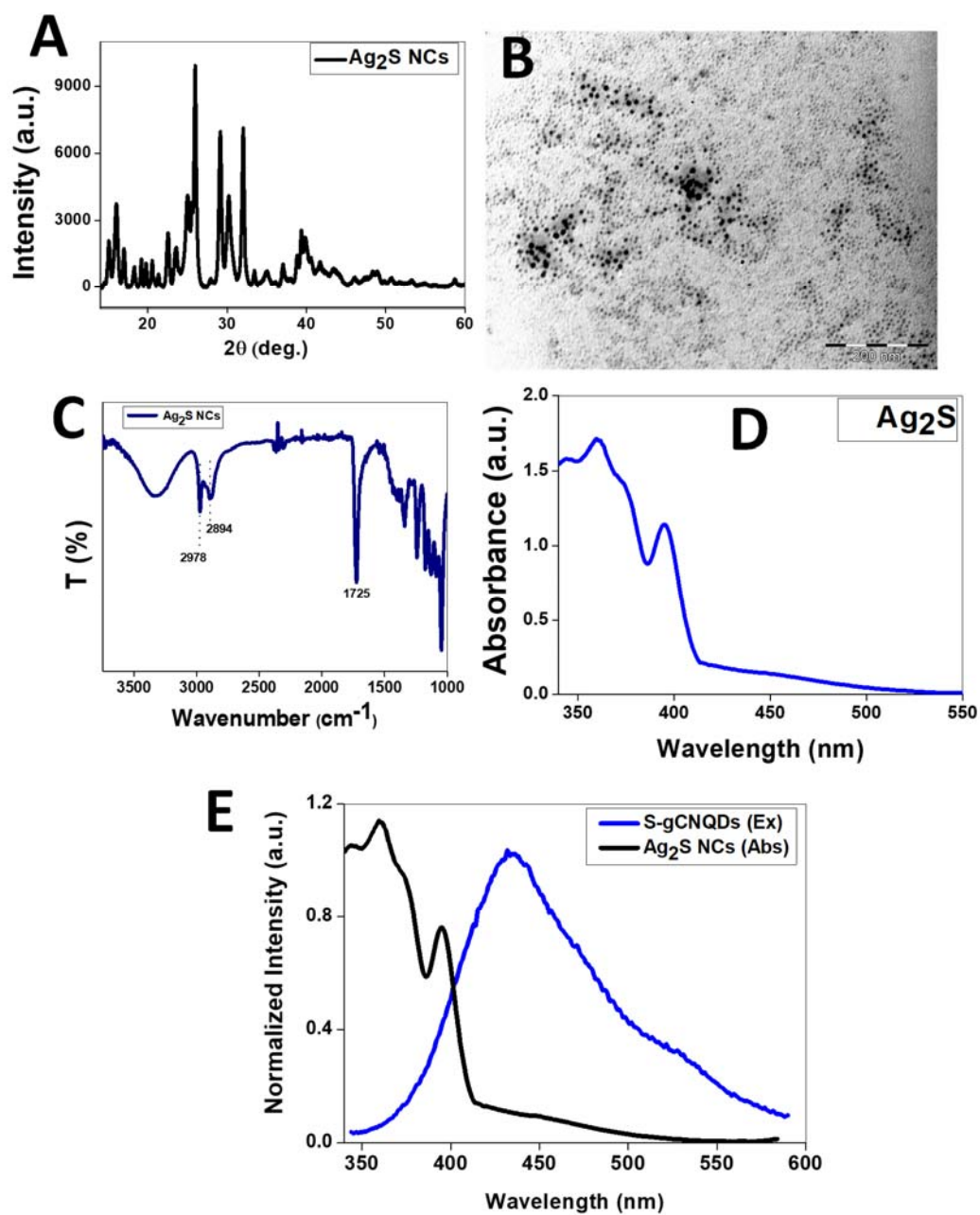
### Synthesis of carboxylic acid-terminated Ag<sub>2</sub>S nanocrystals (NCs)

Water-soluble Ag<sub>2</sub>S NCs were synthesized following reported procedures with some modifications [1]. In a typical procedure, 0.01 g of AgNO<sub>3</sub> salt, 1.2 mL of 3-mercaptopropionic acid and 10 mL of diethylene glycol (DEG) as the reaction solvent, were mixed together and heated up to 120 °C under stirring to form a cloudy mixture. The temperature was further increased to 145 °C for 10 min and the reaction mixture was darkened. The temperature was lowered and the resulting product was allowed to cool to room temperature. Ethanol (95%) was added to the crude product and solid Ag<sub>2</sub>S was collected by centrifugation. The product was purified by centrifugation using a mixture of ethanol and acetone to obtain purified Ag<sub>2</sub>S NCs. The collected solid product was redispersed in water and stored in the refrigerator for further use.

### Characterization Ag<sub>2</sub>S nanocrystals

It is imperative to point out here that the use of 3-MPA was for capping and stabilization of Ag<sub>2</sub>S NCs, and to provide water-solubility with the rich carboxylic groups of 3-MPA. Thus, morphological characterization using X-ray diffraction analysis was carried out for the prepared Ag<sub>2</sub>S NCs. The result verified that the Ag<sub>2</sub>S NCs are crystalline, **Fig. S1A** of ESM. The peaks of diffraction in the XRD pattern are very close to those of the monoclinic Ag<sub>2</sub>S library pattern (JCPDS Card No. 14-0072) [1]. The TEM image of the prepared Ag<sub>2</sub>S NCs revealed particles

which appeared to be separated with non-uniform overall dimension and average size of  $\sim 2.5 \pm 1.1$  nm (**Fig. S1B**). The surface carboxylic acid/sulfide functionalizations were confirmed using FTIR (**Fig. S1C**). Asymmetric and symmetric stretching vibrations are observed for 3-MPA moiety on the  $\text{Ag}_2\text{S}$  NCs. The C–H vibrations of 3-MPA are centered at  $2978\text{ cm}^{-1}$  and  $2894\text{ cm}^{-1}$ . The -COOH group stretching mode appeared as a prominent peak at  $1725\text{ cm}^{-1}$  which is similar to what was observed previously [17]. The characteristic free thiol (SH) absorption (in the MPA) was not observed and this shows that the 3-MPA was bound by the Ag-S binding affinity to the surface of the nanocrystals, which show that the surface of the  $\text{Ag}_2\text{S}$  NCs possesses carboxylic acid terminals due to the use of 3-MPA for capping. The UV-vis absorption spectrum of the  $\text{Ag}_2\text{S}$  NCs displayed absorption band than spans the UV to the visible region, **Fig. S1D**. An intense absorption can be observed at 400 nm which is attributed to the presence of both direct and indirect band gaps in the  $\text{Ag}_2\text{S}$  NCs [1, 2]. In Fig S1E, the overlap of the absorption spectra of  $\text{Ag}_2\text{S}$  NCs and the excitation spectra of S-gCNQDs is shown to reflect the possibility of optoelectronic coupling of the nanoparticles.



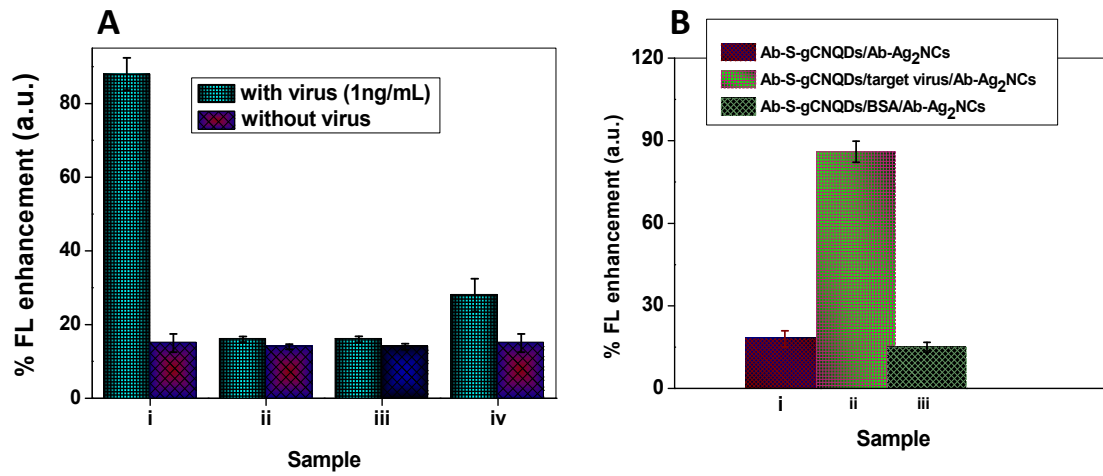
**Fig. S1.** Characterization results of Ag<sub>2</sub>S nanocrystals showing (A) Powder XRD pattern (B) TEM image (C) FTIR spectra and (D) Ground state UV-vis absorption spectra. (E) Overlap of the absorption spectra of Ag<sub>2</sub>S NCs and the excitation spectra of S-gCNQDs.

## Optimization and control studies for the developed immunoassay

The immunoassay experiments were characterized with a series of optimization studies before the generation of calibration plots for optimized sensitivity results. The preliminary optimization included (a) the concentration of antibody conjugated S-gCNQDs or Ag<sub>2</sub>S NCs needed for maximum FL signal; (b) the concentration range of influenza virus that generate maximum FL signal; (c) the incubation/assay duration, and (d) the stability/reproducibility of the detection FL signal of developed system. The following experimental conditions gave optimum performance of the immunoassay detection of influenza virus: (a) 2 mg/mL of Ab-S-gCNQDs and 1 mg/mL of Ab-Ag<sub>2</sub>S NCs in PBS (pH 7.6), respectively. (b) Optimal concentration range of influenza virus for the assay was obtained from 10 fg/mL to 10 ng/mL (c) assay duration was 15 min (d) FL detection signals were stable for more than 15 days. These conditions were further deployed for influenza virus detection in clinical samples.

In order to test the contributions of each species in the developed influenza virus immunoassay protocol, control experiments were conducted. BSA was deployed as a negative control in place of the target influenza virus. The effect of BSA as negative control was tested in the presence of the antibody-conjugated S-gCNQDs or Ag<sub>2</sub>S NCs to demonstrate the specific affinity of the antibody conjugated S-gCNQDs or Ag<sub>2</sub>S NCs to bind to the target virus via antibody-antigen immunoreactions. Another control experiment conducted was to test the interactions between the S-gCNQDs and Ag<sub>2</sub>S NCs with/without antibody

conjugation in the presence or absence of the influenza virus. All experiments were carried under similar conditions. The results obtained from all control experiments are represented in Fig. S2A and B. It can be seen from the results that the FL enhancement of S-gCNQDs was only guaranteed in the presence of the target influenza virus only when the S-gCNQDs and Ag<sub>2</sub>S NCs are conjugated to anti influenza virus A antibody (Fig. S2i). To test the influence of Ag<sub>2</sub>S NCs in the FL enhancement scheme, Ag<sub>2</sub>S NCs (without antibody) and antibody-conjugated Ag<sub>2</sub>S NCs were tested. Only the antibody modified Ag<sub>2</sub>S NCs was involved in the FL enhancement of S-gCNQDs. This is expectedly due to the immunoreactions between the antibody-conjugated nanostructures and the target influenza virus which bring the S-gCNQDs and Ag<sub>2</sub>S NCs close enough to trigger a metal-enhanced fluorescence interaction in the S-gCNQDs. This was also confirmed when no obvious FL change was observed when antibody conjugated Ag<sub>2</sub>S NCs was interacted with S-gCNQDs alone (without antibody) in the presence of the target influenza virus (Fig. S2Aiii). Further, the use of BSA as a control did not result in any change in the FL intensity of S-gCNQDs (Fig. S2Biii) as compared to when the target virus was employed (Fig. S2Bii). These results are proof of the specific contribution of the deployed species to the feasibility of the target virus immunoassay.



**Fig. S2.** (A) Percent change in FL intensity of (i) antibody-S-gCNQDs/antibody-Ag<sub>2</sub>S NCs. (ii) S-gCNQDs/Ag<sub>2</sub>S NCs. (iii) S-gCNQDs/antibody-Ag<sub>2</sub>S NCs. (iv) antibody-S-gCNQDs/Ag<sub>2</sub>S NCs. (B) the FL response of S-gCNQDs with antibody in the presence of Ag<sub>2</sub>S NCS with antibody. (i) Without virus. (ii) With virus and (iii) with BSA.

### Fluorescence quantum yield ( $\Phi_F$ ) determination

The fluorescence quantum yield of the S-gCNQDs was determined by the comparative method using eqn. 1 [3].

$$\Phi_F = \Phi_{F(\text{Std})} \frac{F \cdot A_{\text{Std}} \cdot n^2}{F_{\text{Std}} \cdot A \cdot n_{\text{Std}}^2} \quad (1)$$

Where  $A$  and  $A_{\text{Std}}$  are the absorbances of the sample and the standard at the excitation wavelength, respectively.  $F$  and  $F_{\text{Std}}$  are the areas under the fluorescence curves of the GQDs and the standard, respectively and  $n$  and  $n_{\text{Std}}$  are the refractive indices of the solvent used for the sample and standard, respectively. Rhodamine 6G in ethanol ( $\Phi_F = 0.94$  [4]) was used as the standard.

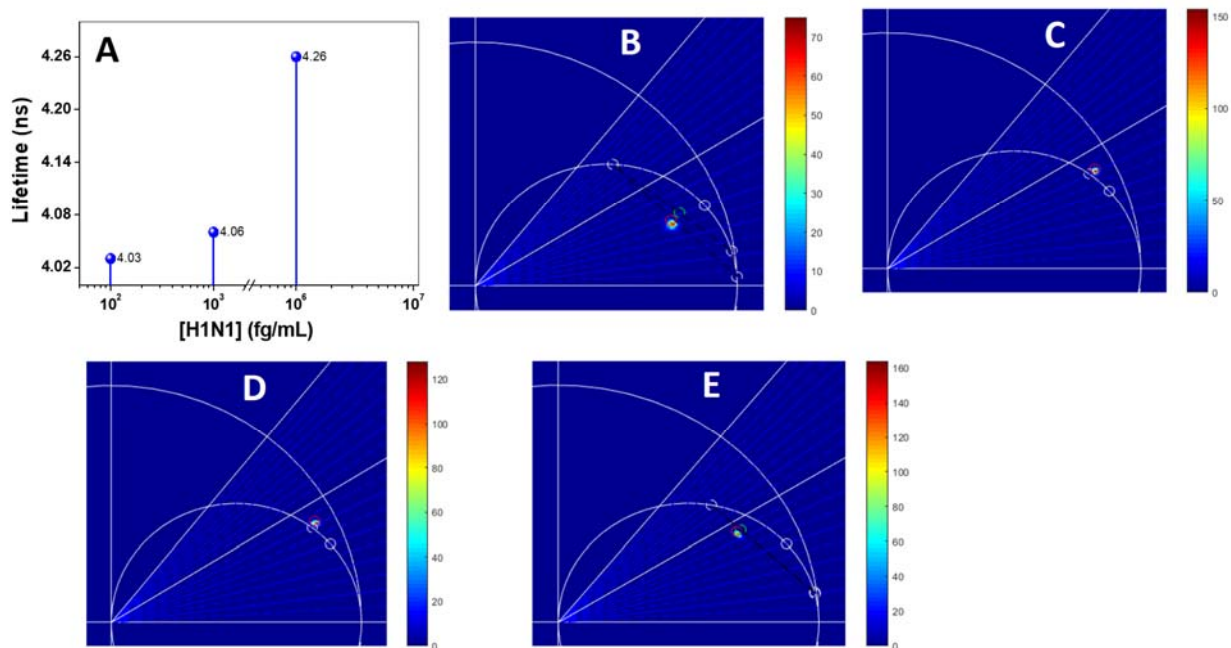
## Fluorescence lifetime imaging microscopy (FLIM) experimental setup and Phasor plot generation

The FLIM images were acquired by the FLIM CMOS camera via the Framelink PCIe card (VCE-CLEX02). The CMOS camera has a pixel array of 128 x 128 pixels; each has a four-tap pixel with a pitch of 22.4  $\mu\text{m}$  x 22.4  $\mu\text{m}$ . The sensor response time is 170ps, measured with 472nm laser diode. The Phasor approach was utilized to determine the fluorescence lifetime of the samples [5], instead of the time delay curve method. The graphical view of the Phasor approach simplifies the analysis of the FLIM images. The base frequency for the measurement was 20 MHz, with 32 phase steps. The FPGA in the FLIM CMOS camera scans 32 phases in a period, which was synchronized with the camera exposure operation. The four non-overlap taps of the pixel resulted in four sets of images over 32 phases in one period of scanning. All four sets of images were rearranged and the images from the same phase were summed to increase the signal to noise ratio. The final single set of 32 phases of images was used for the Phasor analysis. Each point in the Phasor plot corresponds to a pixel in the image. The fluorophore/hybrid nanostructure exhibiting a single component will result in the plot on the semicircle. The plot of a multiple-component fluorophore/nanostructure will be situated inside the semicircle. On the semicircle, a very long lifetime corresponds to the Phasor near the origin, while a very short lifetime corresponds to the Phasor at right side of the semicircle.



## Fluorescence lifetime imaging microscopy (FLIM) of *S*-gCNQDs in the presence of target virus

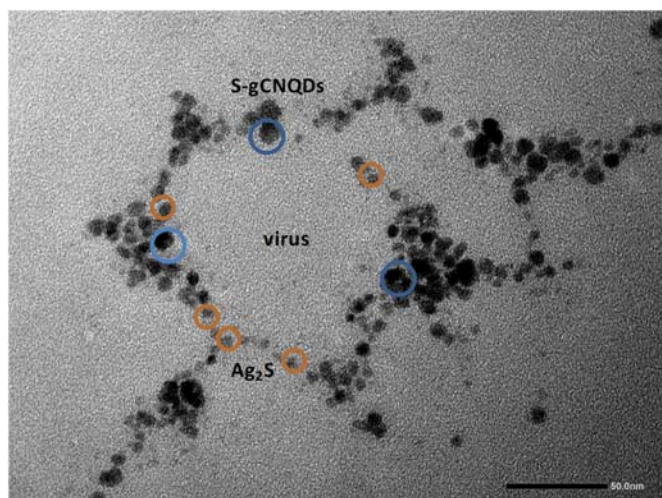
The changes in the lifetime of *S*-gCNQDs as function of the virus amount were studied using FLIM experiment. The frequency-domain FL lifetimes of *S*-gCNQDs when conjugated to the antibody and in the virus-induced nanosandwich system was examined. The FL lifetime of *S*-gCNQDs alone exhibits two-component decay with an average value of 3.19 ns. Meanwhile, in the presence of the H1N1 antibody, the antibody conjugated *S*-gCNQDs reveals a shortening of their average lifetime to 2.66 ns with a Phasor plot, **Fig. S3B**, indicating a multi-component decay. This indicates an interaction between the *S*-gCNQDs and the antibody. The *S*-gCNQDs in the presence of different concentrations of the H1N1 virus at 100 fg/mL, 1.0 pg/mL and 1.0 ng/mL displayed average lifetime values of 4.03, 4.1 and 4.26 ns, respectively (**Fig. S3A, C–E**).



**Fig. S3.** (A) Changes in fluorescence lifetime of *S*-gCNQDs as a function of H1N1 virus concentrations. (B–E) Phasor plots for lifetime of *S*-gCNQDs in the presence (B) Ab (C) 1 ng/mL (D) 1 pg/mL and (E) 1 fg/mL of H1N1 virus.



**Fig. S4.** Commercial RIDT kit for the detection of H3N2 at (A)  $50 \times 10^3$  PFU/mL\* (B)  $10 \times 10^3$  PFU/mL\* (C)  $1 \times 10^3$  PFU/mL\* and (D) 50 PFU/mL. Where A, B, and C denote influenza A virus, influenza B virus, and control, respectively. \* Positive result.



**Fig. S5.** TEM image of immunocomplex of S-gCNQDs/virus/Ag<sub>2</sub>S NCs.

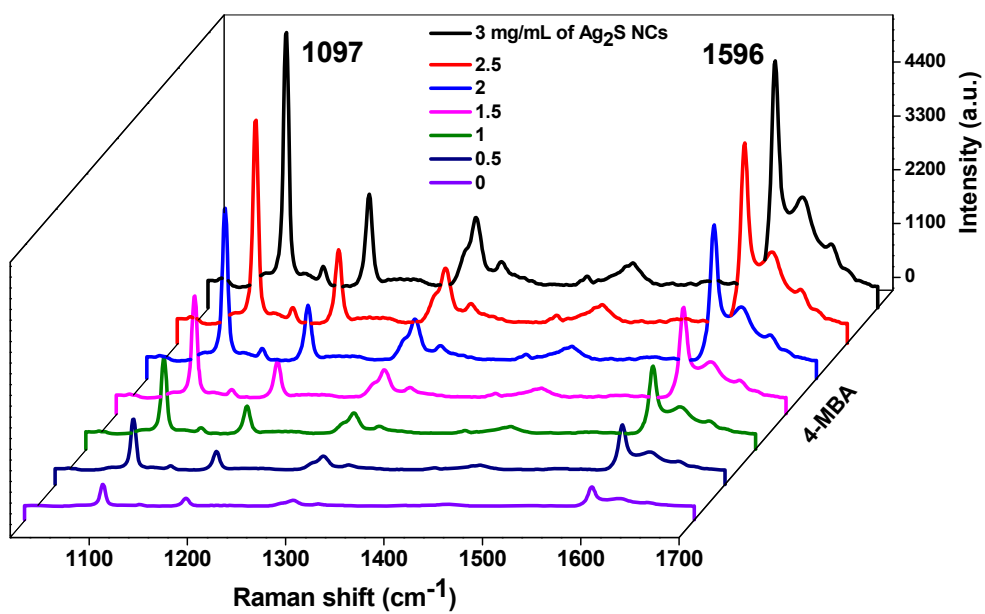
### **Preparation of sample for surface-enhanced Raman scattering (SERS) measurement**

To record the Raman spectra (SERS) of 4-mercaptobenzoic acid (4-MBA) adsorbed on Ag<sub>2</sub>S nanocrystals, 10 μL of 4-MBA (1.0 × 10<sup>-3</sup> M) in methanol mixed with 20 μL of different concentration of Ag<sub>2</sub>S NCs (0.5, 1, 1.5, 2, 2.5 and 3 mg/mL) were dropped onto a silicon substrate, respectively. After air-drying of the solvent, the Raman spectra of the samples were measured. Different laser excitation SERS experiments were conducted using 532 nm laser source. The Raman spectra of S-gCNQDs in the presence of increasing influenza virus concentration and Ag<sub>2</sub>S NCs were also measured using the 532 nm laser source.

### **Demonstration of plasmonic properties of Ag<sub>2</sub>S NCs**

SERS analysis and results have been used as measure of the plasmonic coupling effects of metallic nanostructures by their ability to enhance the optical Raman scattering signals of a Raman reporter in close proximity. In this work and to demonstrate the plasmonic properties of Ag<sub>2</sub>S NCs, SERS analysis was adopted to probe the optical enhancement of the Raman signals of 4-MBA, a classical Raman reporter, in the presence of Ag<sub>2</sub>S NCs. As shown in **Fig. S6**, different concentrations of Ag<sub>2</sub>S NCs were incubated with 4-MBA and the recorded SERS spectra showed significant increase in the Raman peaks at 1097 and 1596 cm<sup>-1</sup>, respectively. The Raman signals of 4-MBA were proportionately enhanced with an enhancement factor in the order of 10<sup>5</sup> compared to 4-MBA alone.

In another SERS experiment, the immunocomplexes formed between the antibody-conjugated-S-gCNQDs and antibody-conjugated-Ag<sub>2</sub>S NCs incubated with 0.1 and 1 ng/mL of influenza virus were also subjected to Raman measurement to test the SERS enhancement of the S-gCNQDs. In Fig. 6B, the spectra contain the characteristic D band at 1349 cm<sup>-1</sup> that arises due to the presence of local structural defects known as sp<sup>3</sup> defects. The other peak at 1599 cm<sup>-1</sup>, called the G-band, a signature of sp<sup>2</sup>-containing graphitic structure of S-gCNQDs. The SERS intensity showed appreciable enhancements of the Raman peaks at 1347 and 1599 cm<sup>-1</sup>, respectively, compared to the antibody-conjugated S-gCNQDs alone (Fig. 6Bi). The SERS analysis results (of S-gCNQDs within the immunocomplex) therefore demonstrates the possibility of metal (plasmonic) coupling effects of Ag<sub>2</sub>S NCs as similarly observed for 4-MBA, but with a lower enhancement factor in the order of 10<sup>2</sup>. The SERS enhancement of S-gCNQDs within the immunocomplex may have been weakened by the intense fluorescence enhancement of S-gCNQDs in the presence of Ag<sub>2</sub>S NCs (Fig. 6A). Meanwhile, no fluorescence signal change or enhancement was observed in the absence of Ag<sub>2</sub>S NCs (Fig. 6A) which further confirms the contribution of Ag<sub>2</sub>S NCs in the detection process.



**Fig. S6.** Raman spectra showing SERS enhancement of 4-MBA in the presence of different concentrations of  $\text{Ag}_2\text{S}$  NCs.

Table S1. A comparison of some recent reports/results on influenza virus detection systems.

Method of detection	Target virus	LOD	Ref.
Fluorescence fiber-optic biosensor	H1N1	13.9 pg/mL	[6]
Magnetofluoro-immunoassay	H1N1	6.07 pg/mL	[7]
Ag-S covalent labelling	H1N1	0.1 pg/mL	[8]
Electrochemical immunosensor	H5N1	2.1 pg/mL	[9]
Metal-enhanced fluoroimmunoassay	H1N1	1 ng/mL	[10]
Peroxidase mimic	H1N1	10 pg/mL	[11]
S-gCNQDs/Ag <sub>2</sub> S NCs assay	H1N1	5.5 fg/mL (DI water ) and 8.48 fg/mL (in serum)	This work



**Table S2.** The recovery/immunoassay of clinically isolated influenza A/Yokohama H3N2 virus using the S-gCNQDs probe. (Detection was carried out in human serum).

Sample	Added (PFU/mL)	Found (PFU/mL)	Recovery (%, n = 3)	*RSD (%)
H3N2 Virus	20 × 10 <sup>3</sup>	19.5 × 10 <sup>3</sup>	97.6±0.15	1.5
	50 × 10 <sup>3</sup>	49 × 10 <sup>3</sup>	98 ±0.20	2.3
	100 × 10 <sup>3</sup>	98 × 10 <sup>3</sup>	98 ±0.27	3.1

\*Relative standard deviation

## References

1. Jiang P, Zhu CN, Zhang ZL, Tian ZQ, Pang DW (2012) Water-soluble Ag<sub>2</sub>S quantum dots for near-infrared fluorescence imaging in vivo. *Biomaterials* 33:5130-5135
2. Li G, Lei Z, Wang QM (2010) Luminescent molecular Ag-S nanoclusters [Ag<sub>62</sub>S<sub>13</sub>(SBU<sup>t</sup>)<sub>32</sub>](BF<sub>4</sub>)<sub>4</sub>. *J Am Chem Soc.* 132:17678-17679
3. Fery-Forgues S, Lavabre D (1999) Are fluorescence quantum yields so tricky to measure? A demonstration using familiar stationary products. *J Chem Ed.* 76:12660-12664.
4. Fischer S, Georges J (1996) Fluorescence quantum yield of Rhodamine 6G in ethanol as a function of concentration using lens spectrometry. *Chemical physics letters.* 260:115-118.
5. Seo MW, Shirakawa Y, Kawata Y, Kagawa K, Yasutomi K, Kawahito S (2018) A time-resolved four-tap lock-in pixel CMOS image sensor for real-time fluorescence lifetime imaging microscopy. *IEEE J. Solid-State Circuits.* 53:2319–2330
6. Chang, Y.F., Wang, S.F., Huang, J.C., Su, L.C., Yao, L., Li, Y.C., Wu, S.C., Chen, Y.M., Hsieh, J.P., Chou, C., 2010. *Biosens. Bioelectron.* 26, 1068–1073.
7. Lee, J., Takemura, K., Park, E.Y., 2018. *Sens. Actuators, B.* 276, 254-261.

8. Li, Y., Hong, M., Qiu, B., Lin, Z., Chen, Y., Cai, Z., Chen, G., 2014. *Biosens. Bioelectron.* 54, 358-364.
9. U. Jarocka, R. Sawicka, A. Góra-Sochacka, A. Sirko, W. Zagórski-Ostoja, J. Radecki, H. Radecka, Electrochemical immunosensor for detection of antibodies against influenza A virus H5N1 in hen serum, *Biosens. Bioelectron.* 55 (2014) 301–306.
10. Ahmed, S.R., Hossain, M.A., Park, J.Y., Kim, S.H., Lee, D., Suzuki, T., Lee, J., Park, E.Y., 2014. *Biosens. Bioelectron.* 58, 33-39.
11. Ahmed, S.R., Kim, J., Suzuki, T., Lee, J., Park, E.Y., 2016. *Biotechnol. Bioeng.* 113, 2298-2303.