Effects of Nitrogen Form on Growth of Muskmelons

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Summary

Muskmelons (*Cucumis melo* L.) were grown in sand with 7 N form treatments $(NO_3: NH_4; 10:0, 8:2, 6:4, 5:5, 4:6, 2:8 and 0:10)$ supplied daily at a concentration of 90 ppm. A total of 78 liters of treatment solution was applied per plant throughout the experiment. The higher NH_4 ratios caused plant growth and fruit fresh weight to decrease, while fruit soluble solids remained slightly higher. With increasing NH_4 ratios N in leaves increased. NO_3 -N and NH_4 -N in the plant parts and medium (sand) reflected the form of N that was supplied. N uptake by plants at all NH_4 ratios was greatest at 5,137 mg per plant. With increasing NH_4 ratios increased, while NO_3 -N and Na, and pH and EC values tended to decrease.

Introduction

Greenhouse muskmelon growers in Shizuoka Prefecture use mainly rapeseed cake and fish cake to supply nitrogen (N) fertilizer in a slowly available form for the production of high quality fruit. In general, rapidly available N in large amounts tends to produce poor quality fruit(8,9). However, when analized for NO₃-N and NH₄-N in soils producing high quality fruit, amounts of these N forms tended to vary with the kind of soils, the number of crops per year, fertilization, cultural practices etc. (8, 9, 10). Thus a question arises as to which muskmelons prefer NO₃-N or NH₄-N per se. The objective of this study was to determine the growth of muskmelons grown in sand applied with nutrient solution containing different ratios of NO₃ : NH₄.

Materials and Methods

Thirty-two uniform seedlings of muskmelons (*Cucumis melo* L. cv. Earl's Favourite Fall No. 1) were used. Treatments consisted of 7 plots receiving $NO_3 : NH_4$ ratios as 10:0, 8:2, 6:4, 5:5, 4:6, 2:8 and 0:10. Control plants were grown in soil fertilized with 7 g N per plant from rapeseed

cake. Thus there were 8 plots, each having 4 replications. On Sept. 14, 1978, seedlings with 2.5 to 3 leaves were planted in 40×40 $\times 20$ cm boxes filled with 19 liters of Tenryu River sand which was classified into Loamy coarse sand (LS). This LS refers to sand for short thereafter. The bottom of boxes was filled with 2 kg rice straw to maintain good aeration. The muskmelons were grown in a three-quarter greenhouse kept at 20°C to 30°C during the day and 18°C to 20°C at night. The N source of NO3 and NH4 was NaNO3 and (NH₄)₂SO₄, respectively. The composition of the nutrient solution except for N was as follows : $Na_2HPO_4 \cdot 12 H_2O$ (1 mM), K_2SO_4 (3 mM), $CaCl_2 \cdot 2 H_2O$ (4 mM), Mg-SO₄·7 H₂O (2 mM), Fe from Fe-EDTA (1 ppm), Mn from MnSO₄ (0.5 ppm), B from $H_{3}BO_{3}$ (0.5 ppm), Zn from ZnSO₄·7 $H_{2}O$ (0.05 ppm), Cu from CuSO₄ · 5 H₂O (0.02 ppm) and Mo from $Na_2MoO_4 \cdot 2H_2O$ (0.05) ppm). Nutrient solutions containing 90 ppm N (NO_3+NH_4) at pH 6.0 were applied 1 or 2 times each day in equal quantities of 0.75 liters to each box whether it was cloudy or sunny. No solutions were applied on rainy days. A total of 78 liters of solution was applied per plant throughout the experiment. The control plot was basically fertilized with 4 g N from rapeseed cake, 10 g P_2O_5 from calcium superphosphate, 8 g K₂O from K₂SO₄

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and 20 g CaO from Ca(OH)₂. In addition, 1.5 g N from rapeseed cake were applied on Oct. 4 and Oct. 27, 1978. Control plants were watered at the same time the treated plants received nutrient solutions. Total tissue N determination followed the method used by Peterson and Chesters(13). Tissue NO₃-N and NH₄-N were determined electrometically, using a distilled water extract of dried, ground plant tissue. Tissue K, Na, Ca and Mg were determined by an atomic absorption photometer, and P by an ammonium molybdate procedure. Samples of sand mediums were taken on the 29 th, 57 th and 83 rd day after planting, and air-dried, and NH₄-N and NO₃-N were determined. Also the chemical properties of the soil solution were determined at the end of the experiment on the extract centrifugally taken at pF 0 to 3.8. The resulting data were subjected to the analysis of variance, and differences among means were tested by Duncan's method.

Results

1. Plant and fruit growth (Table 1).

Dry weight of the leaves, stem and whole plant was greatest at $NO_3 : NH_4=8:2$, and that of leaves, stem and roots was less at 2:8, and 0:10. Dry weight of the roots was greater at $NO_3 : NH_4=6:4$, 5:5, and 4:6, while less at 10:0, and 2:8. Fruit fresh weight tended to decrease with de creasing ratios of $NO_3 : NH_4$ except for $NO_3 : NH_4=8:2$. Fruit soluble solids were slightly greater at higher NH_4 ratios. Fruit external appearance such as shape, net and rind color, and internal quality such as flesh texture and color, except for soluble solids, were not affected by treatments. The control fruit fresh weight was slightly greater, while soluble solids were slightly lower.

2. Elemental content in leaves (Table 2).

With decreasing ratios of NO₃: NH₄, N tended to increase, while P, Na, Ca and Mg tended to decrease. The control Ca was greatest, while N, K and Na was least.

 NO₃-N and NH₄-N content in plant parts, and N plant uptake (Table 3).

 NO_3 -N in plant parts decreased with decreasing ratios of NO_3 : NH_4 . At the NO_3 : $NH_4=10:0$ NO_3 -N was highest in the stem followed by the roots and then the leaves. The control NO_3 -N was lower in the stem. In leaves NH_4 -N was lower than NO_3 -N, but was higher at $NO_3: NH_4=4:6$, 2:8, and 0:10 than the others. NH_4 -N in the roots tended to increase with decreasing ratios of $NO_3: NH_4$. The N plant uptake (total of leaves, stem, roots and fruit) was greatest at all NH_4 ratios showing 5, 137 mg per plant.

4. NO_3 -N and NH_4 -N content in sand during growing period (Table 4).

 NO_3 -N in sand tended to increase with increasing ratios of NO_3 : NH_4 , and tended to increase with time at each ratio of NO_3 : NH_4 . NH_4 -N in sand tended to increase with decreasing ratios of NO_3 : NH_4 with increasing time. The control NH_4 -N was as low as 4.2 ppm on Dec. 6.

5. Chemical properties of soil solution

NG 197	Plant height	Leaf dry	Stem dry	Root dry	Whole plant	Fruit	
NO₃ : NH₄	at harvest(cm)	wt (g)	wt (g)	wt (g)	dry wt (g)	Fresh wt (g)	Soluble solids (%)
10: 0	116 ^{a z}	64. 9 ^{a b}	19.0 ^{ab}	2. 4 ^b	199 ^{abc}	1121ª	14.1°
8: 2	116ª	70.2ª	21.0ª	4.3 ^{ab}	213ª	936 ^b °	14.4 ^{bc}
6:4	120ª	58.8 ^{ab}	18.4 ^{abc}	5. 5ª	207 ^{ab}	1030 ^{a b}	15.0 ^{ab}
5:5	118ª	64. 5 ^{a b}	19.3ªb	5.4ª	205 ^{a b}	923 ^{b c}	14.9 ^{ab}
4:6	112ª	55.1 ^{ab}	16.6 ^{bcd}	5. 6ª	186 ^{bcd}	907 ^{bc}	15.6ª
2: 8	111ª	47.6 ^b	14.0 ^d	2.5 ^b	175° d	910 ^{bc}	15.5ª
0:10	112ª	50.7 ^{a b}	14. 8 ^{c d}	3. 2 ^{a b}	168 ^d	832°	15.7ª
Rs C ^y	114	53.0	18.8	2.3	209	1195	14.3

Table 1. Effect of nitrogen form on the growth of muskmelons in sand culture.

^zMean separation in columns by Duncan's multiple range test, 5% level. ^yRapeseed cake.

NO3 : NH4	Ν	Р	К	Na	Ca	Mg
10: 0	2.98° ^z	0.65ª	2. 49 ^b	0.27ª	6.75ª	1.69ª
8: 2	3.11°	0.56 ^b	2.97 ^{ab}	0.20 ^b	5.20 ^b	1.26 ^b
6:4	3.54 ^{bc}	0.57 ^b	3.02 ^{ab}	0.20 ^b	4.24°	1.11°
5:5	3.35 ^{bc}	0. 49°	3.26ª	0.16 ^b	3. 86°	1.06°
4:6	3.33 ^{b c}	0.49°	3.26ª	0.18 ^b	3.98°	1.11°
2: 8	3.85ªb	0.46°	3.08 ^{ab}	0.11°	3. 48 ^{cd}	0.87ª
0:10	4. 48 ^a	0. 41 ^d	3.05 ^{ab}	0.10°	3. 31 ^d	0.88ª
Rs C ^y	2.08	0.44	2.16	0. 07	7.15	0.73

 Table 2. Effect of nitrogen form on the elemental content of muskmelon leaves grown in sand culture (% of dry wt).

^{z,y}Same as Table 1.

Table 3. Effect of nitrogen form on the nitrate and ammonium content of plant parts and N amount taken up by muskmelon plants grown in sand culture.

NO3:NH4	NO ₃ -N	V (ppm of d	ry wt)	NH4-1	N taken up ^x		
	Leaves	Stem	Roots	Leaves	Stem	Roots	(mg/plant)
10: 0	773 ^{a z}	3423ª	1061ª	36 ^b	284ª	267°	4446
8:2	633 ^{a b}	2755ª	661 ^b	33ъ	255ª	275°	4470
6:4	645 ^{ab}	1605 ^b	385°	35ъ	371ª	315 ^b	4848
5:5	609 ^{a b}	818°	222ª	36 ^b	308ª	308 ^b	4557
4:6	481 ^{b c}	583°	123°	47 ^{ab}	391ª	332ªb	4425
2:8	306°	260°	44 ^f	55ª	331ª	354ª	4412
0:10	337°	178°	31 ^f	54ª	306ª	343ª	5137
Rs C ^y	539	213	218	22	311	321	2949

^{z,y}Same as Table 1. *Not subjected to statistical analysis.

		NO ₃ -N		NH4-N				
$NO_3: NH_4$	Oct. 13	Nov. 11	Dec. 6	Oct. 13	Nov. 11	Dec. 6		
10:0	4.4 ^{az}	7.5 ^{ab}	22.0ª	1.8 ^d	1.4 ^e	2. 3 ^g		
8:2	2.9 ^b	8.7ª	14.4 ^b	1.7 ^d	1.4 ^e	6.7 ^f		
6:4	3.0ªb	6.6 ^b	9.3°	1.5ª	1.9 ^{d e}	12.8°		
5:5	2.6 ^b	4. 3°	7.4 ^{cd}	1.6ª	2.1 ^d	16.7ª		
4:6	3. 2 ^{ab}	4. 4°	6.1 ^{de}	3.0°	3. 3°	20.0°		
2: 8	2. 3 ^b	2.8°	4. 2 ^{e f}	5.4 ^b	5.7 ^b	31. 3 ^b		
0:10	1.8 ^b	1.0 ^d	2.0 ^f	6. 3ª	7.7ª	35.7ª		
Rs C ^y	51.7	7.6	21.8	4.5	4.0	4.2		

Table 4. Nitrate and ammonium content of sand during growing period (ppm of dry sand).

(Table 5).

With increasing ratios of NO₃ : NH₄, NO₃-N and Na, and pH and EC values increased, while NH₄-N decreased. P was higher at NO₃ : NH₄=10 : 0, and K, Ca and Mg was lower at NO₃ : NH₄=0 : 10. The control pH was as high as 6.73, while NH₄-N, K, Na and Mg were markedly low.

Discussion

This experiment showed that higher NH_4 ratios reduce plant growth and fruit weight, increase fruit soluble solids, decrease Ca and Mg in leaves, and increase N in leaves of muskmelons in sand culture. Similar results were reported for some vegetable crops (1, 2, 3, 4, 5, 6, 7, 12, 15). The reason of reduced

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NO3 : NH4	NO ₃ -N (ppm)	NH₄-N (ppm)	P (ppm)	K (me/l)	Na (me/l)	Ca (me/l)	Mg (me/l)	pH	EC (m℧/cm)
10: 0	108°z	5 s	9.7ª	10. 3ª	34.7ª	14. 1 ^{ab}	13.0 ^{ab}	6.68ª	7.18ª
8: 2	86 ^b	14 ^f	4.3 ^b	9.7 ^{ab}	23. 4 ^b	13.2 ^{abc}	11.5 ^{ab}	6.00 ^b	6.88 ^{a b}
6:4	55°	29°	3.4 ^b	10. 3ª	21.6 ^b	14.5 ^{ab}	13.9ª	5. 20°	6.96 ^{ab}
5:5	37 ^d	38ª	2.8 ^b	9.9ªb	19.1 ^{bc}	14.7ª	13.8 ^{ab}	5.00 ^{cd}	6.06 ^{bc}
4:6	35ª e	48°	2.7 ^b	9.7ªb	16.1°	15.1ª	14.3ª	4.80 ^{d e}	5. 89 ^{c d}
2:8	24 ^e	73 ^b	3.1 ^b	7.5 ^{bc}	10. 5 ^d	12.4 ^{bc}	12.0 ^{ab}	4. 59 ^{e f}	4. 93 ^{d e}
0:10	8 ^e	86ª	3.7 ^b	5.7°	5.6°	11.8°	10.6 ^b	4.50 f	4. 32 ^e
Rs C ^y	98	4	5.7	3. 1	6.5	27.7	8.1	6.73	4.33

Table 5. Chemical properties of soil solution $(pF=0\sim3.8)$ at the end of experiment.

^{z,y}Same as Table 1.

plant growth and fruit weight (Table 1) may be due to excess NH4 uptake by plant, depressed Ca absorption by plants, a lowering of medium pH etc. In this experiment NH4 injury was not found on leaves. However, NH₄-N content in sand and soil solution seemed to be high enough to reduce plant growth and fruit weight as showing 35.7 and 86 ppm at NO_3 : $NH_4 = 0$: 10, respectively. Fruit soluble solids were slightly greater at higher NH₄ ratios. This seems to be due to fruit weight because small fruits have generally higher soluble solids under greenhouse cultivation. In general, higher NH4 supply promotes protein synthesis in plants. However, vigorous protein synthesis often reduces plant growth due to an excess NH4 when the carbohydrate is deficient within plants. As shown in Table 2, N content in leaves increased with increasing NH4 ratios. In comparing these results to another study (11), more than 3.5% N in leaves seem to be high for plant growth. Thus the ratio of NO₃: NH₄ for optimum plant growth seems to be 10:0 or 8:2 judging only from the N content in the leaves.

According to results by some reporters(3, 4, 12, 15,), the symptom of NH_4 injury found on leaves was rolling, and marginal and interveinal chlorosis or necrosis. In this experiment any symptom of NH_4 injury was not found on leaves although growth was reduced, and leaves became darker green at higher NH_4 ratios. The reason seems to be related to cultivation time. Iwata reported that NH_4 injury appears more rapidly on summer crops than winter crops(3). Another study of ours showed that NH_4 injury appears on leaves as marginal chlorosis and necrosis in muskmelons grown in early summer, and that NH_4 -N content in leaves is much higher than that in this experiment (11).

Ca and Mg in leaves decreased with increasing NH₄ ratios (Table 2). These reductions can be explained by the electrochemical equilibrium theory that NH₄ uptake must be accompanied by either inorganic anion uptake and/or higher organic anion production, or reduced uptake of inorganic cations(14). As to Ca nutrition we have pointed out that muskmelons are luxurious consumers of this element(10). Considering this, the decreasing Ca absorption with higher NH₄ ratios is highly correlated with the poor growth of plants.

Nutrient solutions with higher NH₄ ratios became extremely acid as the plants extracted NH₄⁺ ions from the soil solution (Table 5). This increased acidity seems to be associated with poor utilization of NH₄, Ca and Mg which restricted plant growth. Maynard *et al.* reports that the growth of some vegetable crops is increased by maintaining sand medium pH near neutrality by the addition of CaCO₃ regardless of N source(12). Similar results with muskmelons will be published separately.

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窒素の形態がメロンの生育に及ぼす影響

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摘 要

窒素 (NO₃ と NH₄ の合計) 濃度 90 ppm で, NO₃: NH₄ の比率 10:0, 8:2, 6:4, 5:5, 4:6, 2:8, 0:10 の7 処理を設け、メロンを砂耕栽培した.全生育期間中 の1 株当たりの処理溶液は78*l* であった.NH₄の比率が 高くなると生育は低下し、果重は減少したが、果実の糖 度はやや増加した.NH₄の比が増加するにつれ、葉のN 含量は増加した. 植物体各部及び培地(砂)の NO_3 -N と NH₄-N 含量は, 施用した窒素の形態をよく反映した. 1 株当たりの窒素吸収量は, NH₄のみを施用した区が 5,137 mg で最も多かった. NH₄の比が増加するにつれ, 土壌溶液中の NH_4 -N 含量は増加したが, NO_3 -N, Na 含 量, pH, EC は減少する傾向がみられた.