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Graphical Abstract





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Thapsigargin-induced ER stress suppressive compounds from the mushroom *Mycoleptodonoides aitchisonii*

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Endoplasmic reticulum (ER) is an important organelle for the synthesis, correct folding, post-translation modification and transport of nascent proteins to different destinies.^{1,2} ER stress is caused by disturbances in the structure and function of the ER with the accumulation of misfolded proteins and alterations in calcium homeostasis. In case of prolonged or aggravated ER stress, cellular signals leading to cell death are activated³. ER stress has been reported to cause not only neurodegenerative diseases³⁻⁵ but also some other diseases, such as diabetes, atherosclerosis, or heart and liver disease.⁶

Therefore, protective activity against ER stress is possibly an important target for addressing these diseases, and the demand for new lead compounds prompted us to screen the protective activity of mushroom extracts. During the screening for ER stress protecting effects of the extracts of various mushrooms, we previously reported the protective compounds from the mushrooms *Hericium erinaceus*,⁷⁻⁹ *Termitomyces titanicus*,¹⁰ and *Leccinum extremiorientale*.¹¹ Moreover, γ -lactones and phenylpentanols having protective activity against toxicity of tunicamycin (TM) or thapsigargin (TG) from the mushroom *Mycoleptodonoides aitchisonii* have been reported.^{12,13} In the course of our continuing search for ER stress protecting

ABSTRACT

Two novel compounds 1 and 2, along with two known ones (3 and 4) were isolated from the edible mushroom *Mycoleptodonoides aitchisonii*. The structures of 1-4 were determined and identified by interpreting their spectroscopic data. All the compounds showed protective activity against endoplasmic reticulum stress-dependent cell death.

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compounds from the mushroom M. *aitchisonii*, we found active compounds. We report here the isolation and structure determination of novel compounds (1 and 2), together with two known ones (3 and 4).



The fresh fruiting bodies of *M. aitchisonii* were extracted with EtOH and then with acetone. After the solutions had been combined and concentrated under reduced pressure, the concentrate was partitioned between CH_2Cl_2 and H_2O and then between EtOAc and H_2O . Compounds **1**, **3**, and **4** were obtained

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from the CH_2Cl_2 -soluble part and compound **2** was purified from the EtOAc-soluble part.¹⁴

Compound 1 was isolated as a colorless oil. The IR spectrum showed absorption bands at 2929 cm⁻¹ and 1735 cm^{-1,15} Its molecular formula was determined as C13H26O4 by HRESIMS $(m/z \ 269.1730 \ [M+Na]^+; \ calcd. \ for \ C_{13}H_{26}NaO_4, \ 269.1729),$ indicating the presence of one degree of unsaturation in the molecule. The complete assignment of all the protons and carbons was accomplished by HMOC, COSY, and HMBC experiments as shown in Table 1 and Figure 1. The DEPT experiment and the molecular formula indicated the presence of three methyls, eight methylenes, a methine, and a quaternary carbon. The ethoxycarbonylpropyl moiety was elucidated by the COSY correlations (bold line in Fig. 1) and the HMBC correlations (H-1'/C-2', H-1'/C-1, H-2'/C-1', H-2/C-1, H-2/C-3, H-3/C-1, H-3/C-2) and the chemical shift of C-1 (δ_C 173.8). The dimethoxypropanyl group was suggested by the NMR signals [$\delta_{\rm H}$ 3.29 (6H, s) and δ_c 52.6], the HMBC correlations [H-7/C-8, H-7/C-9, H-8/C-7, H-8/C-9, H-9/C-7, H-9/C-8, H-9/MeO-, MeO-/C-9], and the COSY correlations (H-7/H-8, H-8/H-9)(Fig. 1). of the ethoxycarbonylpropyl, The connection the dimethoxypropanyl and the other part was suggested by the molecular formula in 1. All the data allowed us to conclude that 1 was ethyl 9,9-dimethoxynonanoate.



Figure 1. COSY and HMBC correlations of 1 and 2.

1 (in CDCl ₃)				2 (in CD ₃ OD)	
position	$\delta_{\rm H}$ (mult., J in Hz)	$\delta_{ m C}$	position	$\delta_{ m H}$ (mult., J in Hz)	$\delta_{\rm C}$
1		173.8	1		
2	2.26 (t, 7.4)	34.4	2	3.64 (m)	77.4
3	1.58 (m)	24.9	3	3.60 (m)	73.2
4, 5, 6	1.29 (m)	29.0, 29.2	4	1.66 (m), 1.92 (m)	36.1
		29.3	5	3.71 (m), 3.77 (m)	60.4
7	1.29 (m)	24.5	1'	2.61 (dd, 14.0, 8.9)	40.4
8	1.55 (m)	32.5		2.97 (dd, 14.0, 3.4)	
9	4.32 (t, 5.7)	104.6	1''		140.9
MeO-9	3.29 (s)	52.6	2", 6"	7.24 (m)	130.5
1'	4.10 (q,7.3)	60.1	3", 5"	7.24 (m)	129.2
2'	1.23 (t. 7.3)	14.2	4''	7.16 (m)	127.0

Table 1. ¹H and ¹³C NMR data for 1 and 2

Compound 2 was isolated as a colorless crystal showing a specific rotation of $[\alpha]_D^{28}$ +15 (c 0.25, CHCl₃) with a melting point of 58-60°C. The IR spectrum showed absorption bands at 3360 cm⁻¹, 2916 cm⁻¹, and 1454 cm⁻¹. Its molecular formula was determined as C₁₁H₁₄O₂ by HRESIMS (*m*/*z* 201.0921 [M+Na]⁺; calcd. for C₁₁H₁₄NaO₂, 201.0892), indicating the presence of five degrees of unsaturation in the molecule. The complete assignment of all the protons and carbons was accomplished by

DEPT, HMQC, COSY, and HMBC experiments as shown in Table 1 and Figure 1. The DEPT experiment and the molecular formula indicated the presence of three methylenes, seven methines, and a quaternary carbon. The presence of the benzyl group was suggested by the COSY correlations (bold line in Fig. 1) and the HMBC correlations (H-1'/C-1", H-1'/C-2", H-2"/C-1', H-2"/C-1", H-2"/C-3", H-2"/C-4", H-3"/C-2", H-3"/C-4", H-4"/C-1", H-4"/C-2", H-4"/C-3"). The tetrahydrofuran moiety was elucidated by the COSY correlations (H-2/H-3, H-3/H-4, H-4/H-5) and the HMBC correlations (H-2/C-3, H-2/C-4, H-3/C-2, H-3/C-4, H-3/C-5, H-4/C-2, H-4/C-3, H-4/C-5, H-5/C-2, H-5/C-4) and the chemical shifts of C-2 (δ_C 77.4) and C-5 (δ_C 60.4). The connection between the benzyl and tetrahydrofuran moieties was determined by the HMBC correlations (H-1'/C-2, H-1'/C-3, H-2/C-1', H-2/C-1", H-3/C-1'). The presence of the hydroxy group at C-3 was suggested by the molecular formula and the chemical shift ($\delta_{\rm H}$ 3.60 and $\delta_{\rm C}$ 73.2). All the data allowed us to conclude that 2 was 2-benzyltetrahydrofuran-3-ol. Although NOESY data of 2 did not give significant information about its stereochemistry, the NMR data of 2 were completely different from those of $(2R^*,$ $3R^*$)-2-benzyltetrahydrofuran-3-ol that has been reported as a synthetic compound.^{16,17} Therefore, relative configuration of 2 was determined to be $(2S^*, 3R^*)$.

By comparison of the NMR data and mass spectra of compounds **3** and **4** with the previous reported papers, they were identified as shown in Figure 1. Compound **3** has been reported as an intermediate in the synthesis of cyclopentapyranone, but this is the first report of isolation of **3** from a natural source.¹⁸ Its stereochemistry remains undetermined. Compound **4** has been isolated as an inducer of NAD(P)H:quinone oxidoreductase 1 from this mushroom.¹⁹



Figure 2. Protective activity of **1–4** against ER stress-dependent cell death. Neuro2a cells were incubated with various concentrations of compounds in the absence or presence of 10 nM thapsigargin (TG) for 24 h. The cell viabilities were analyzed by MTT assay, and the values were represented as the mean \pm SE of the relative percentage of surviving cells compared with the untreated cells (n=10–12). (*) P<0.01, Tukey-Kramer multiple comparisons tests.

Compounds 1–4 were subjected to the protective activity assay against ER stress-dependent cell death caused by TM or TG.¹³ ER stress was induced by addition of TM or TG into the culture medium of Neuro2a cells in the presence or absence of each compound. TM is an inhibitor of *N*-linked glycosylation and the formation of *N*-glycosidic protein-carbohydrate linkages.²⁰ TG, an inhibitor of sarcoplasmic/endoplasmic reticulum Ca²⁺- ATPase, also induces ER stress by disrupting the homeostatic balance of the Ca²⁺ concentration in the ER.²¹ All the compounds did not show the protective activity against TM-toxicity (data not shown). On the other hand, in the assay using TG, all the compounds showed the protective activity at more than 40 μ M (Fig. 2).

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References and notes

- Rao, R. V.; Ellerby, H. M.; Bredesen, D. E., Cell Death Differ., 2004, 11, 372–380.
- Breckenridge, D. G.; Germain, M.; Mathai, J. P.; Nguyen, M.; Shore, G. C. *Oncogene* 2003, 22, 8608–8618.
- Lindholm, D.; Wootz, H.; Korhonen, L. Cell Death Differ. 2006, 13, 385–392.
- Katayama, T.; Imaizumi, K.; Sato, N.; Miyoshi, K.; Kudo, T.; Hitomi, J.; Morihara, T.; Yoneda, T.; Gomi, F.; Mori, Y.; Nakano, Y.; Takeda, J.; Tsuda, T.; Itoyama, Y.; Murayama, O.; Takashima, A.; St George-Hyslop, P.; Takeda, M.; Tohyama, M. *Nat. Cell. Biol.* 1999, *1*, 479–485.
- Tamatani, M.; Matsuyama, T.; Yamaguchi, A.; Mitsuda, N.; Tsukamoto, Y.; Taniguchi, M.; Che, Y. H.; Ozawa, K.; Hori, O.; Nishimura, H.; Yamashita, A.; Okabe, M.; Yanagi, H.; Stern, D. M.; Ogawa, S.; Tohyama, M. *Nat. Med.* **2001**, *7*, 317–323.
- 6. Yoshida, H. FEBS J. 2007, 274, 630–658.
- Nagai, K.; Chiba, A.; Nishino, T.; Kubota, T.; Kawagishi, H. J. Nutr. Biochem. 2006, 17, 525–530.
- Ueda, K.; Tsujimori, M.; Kodani, S.; Chiba, A.; Kubo, M.; Masuno, K.; Sekiya, A.; Nagai, K.; Kawagishi, H. *Bioorg. Med. Chem.* 2008, *16*, 9467–9470.
- Ueda, K.; Kodani, S.; Kubo, M.; Masuno, K.; Sekiya, A.; Nagai, K.; Kawagishi, H. *Biosci. Biotechnol. Biochem.* 2009, 73, 1908– 1910.
- Choi, J.-H.; Maeda, K.; Nagai, K.; Harada, E.; Kawada, M.; Hirai, H.; Kawagishi, H. Org. Lett. 2010, 12, 5012–5015.
- 11. Choi, J.-H.; Ozawa, N.; Yamakawa, Y.; Nagai, K.; Hirai, H.; Kawagishi, H. *Tetrahedron*, **2011**, *67*, 6649–6653.
- Choi, J.-H.; Horikawa, M.; Okumura, H.; Kodani, S.; Nagai, K.; Hashizume, D.; Koshino, H.; Kawagishi, H. *Tetrahedron*, 2009, 65, 221–224.
- Choi, J.-H.; Suzuki, T.; Okumura, H.; Noguchi, K.; Kondo, M.; Nagai, K.; Hirai, H.; Kawagishi, H. J. Nat. Prod. 2014, 77, 1729– 1733.
- 14. The material, mature fruiting bodies of M. aitchisonii were collected at Narusawa village, Yamanashi Prefecture in Japan, in August 2006. The fresh fruiting bodies of M. aitchisonii (13.0 kg) were extracted with EtOH (21 L, three times) and then acetone (10 L. once). After the solutions had been combined and concentrated under reduced pressure, the concentrate was partitioned between CH₂Cl₂ and H₂O and then between EtOAc and H₂O. The CH₂Cl₂soluble part (73.6 g) was fractionated by silica gel flash column chromatography (hexane/CH2Cl2 7:3, hexane/EtOAc 9:1, 5:5, CH₂Cl₂/MeOH 8:2, 5:5, 0:10, 2 L each) to obtain 25 fractions (fractions 1-25), and fraction 19 (6.9 g) was further separated by silica gel flash column chromatography (CH2Cl2/acetone 9:1, 7:3, 3:7, acetone/MeOH 5:5, and MeOH, 1 L each) and 24 fractions were obtained (fractions 19-1 to 19-24). Fraction 19-15 (751 mg) was separated by silica gel flash column chromatography (CH₂Cl₂, CH₂Cl₂/MeOH 99:1, 95:5, 9:1, 1:1, and MeOH, 1 L each) to obtain 11 fractions (fractions 19-15-1 to 19-15-11), and then fraction 19-15-6 (92.7 mg) was further separated by normal-phase HPLC (Senshu Pak AQ, CHCl₃/MeOH 98:2) to obtain 15 fractions (fractions 19-15-6-1 to 19-15-6-15). Fraction 19-15-6-12 (5.6 mg) was separated by reverse-phase HPLC (Develosil C30-UG-5, 30% MeOH) to afford compound 3 (1.7 mg). Fraction 7 (13.1 g) was separated by silica gel flash column chromatography (CH_2Cl_2, CH_2Cl_2/MeOH 99:1, 98:2, 95:5, 5:5 and MeOH, 1 L each), and 10 fractions were obtained (fractions 7-1 to 7-10).

Compound 1 (3.0 mg) was purified by normal-phase HPLC (Senshu Pak AQ, hexane/CHCl₃1:1) from fraction 7-2 (51 mg). Fraction 18 (374 mg) was separated by preparative silica TLC (CH₂Cl₂/MeOH 9:1) to obtain 6 fractions (fractions 18-1 to 18-6) and then fraction 18-5 (72 mg) was further separated by preparative ODS TLC (90% MeOH) to afford compound 4 (15.2 mg). Similarly, the EtOAc-soluble part (25.0 g) was separated by silica gel flash column chromatography (CH₂Cl₂, CH₂Cl₂/MeOH 9:1, 7:3, 5:5, and MeOH, 1 L each), and 10 fractions were obtained (fractions 1 to 10). Fraction 7 (58.7 mg) was further separated by normal-phase HPLC (Senshu Pak AQ, hexane/CHCl₃ 96:4) to afford compound 2 (2.5 mg).

- 15. ¹H NMR spectra (one- and two-dimensional) were recorded on a Jeol lambda-500 spectrometer at 500 MHz, while ¹³C NMR spectra were recorded on the same instrument at 125 MHz. The HRESIMS spectra were measured on a JMS-T100LC mass spectrometer. A Jasco grating infrared spectrophotometer was used to record the IR spectra. The specific rotation values were measured by using a Jasco DIP-1000 polarimeter. Melting points were measured on a J-Science RFS-10 hot stage microscope. HPLC separations were performed with a Jasco Gulliver system using a reverse-phase HPLC column (Develosil C30-UG-5, Nomura chemical, Japan) and a normal-phase HPLC column (Senshu Pak AQ, Senshu scientific Co., Ltd., Japan). Silica gel plate (Merck F254), ODS gel plate (Merck F254) and silica gel 60 N (Kanto Chemical, Japan) were used for analytical TLC and for flash column chromatography, respectively.
- 16. Bach, T. Liebigs Ann. 1995, 1995, 1045-1053.
- Compound 2: ¹³C NMR (CDCl₃); & 75.2 (C-2), 74.0 (C-3), 33.1 (C-4), 61.5 (C-5), 38.4 (C-1'), 138.1 (C-1''), 129.3 (C-2'', C-6''), 128.7 (C-3'', C-5''), 126.6 (C-4''), (2*R**, 3*R**)-2-benzyltetrahydrofuran-3-ol: ¹³C NMR (CDCl₃); & 83.9 (C-2), 72.0 (C-3), 35.1 (C-4), 65.7 (C-5), 35.5 (C-1'), 138.6 (C-1''), 129.1 (C-2'', C-6''), 128.4 (C-3'', C-5''), 126.2 (C-4'').
- 18. Ohba, M.; Haneishi, T.; Fujii, T. Chem. Pharm. Bull. 1996, 44, 525–529.
- Kokubo, T.; Taniguchi, Y.; Kanayama, M.; Shimura, M.; Konishi, Y.; Kawagishi, H.; Yamamoto, M.; Shindo, K.; Yoshida, A. Food Chem. 2011, 129, 92–99.
- 20. Mahoney, W. C.; Duksin, D. J. Biol. Chem. 1979, 254, 6572-6576.
- Hitomi, J.; Katayama, T.; Taniguchi, M.; Honda, A.; Imaizumi, K.; Tohyama, M. *Neurosci. Lett.* 2004, 357, 127–130.